

# 3-methyl-1,7-octadiene

<b>Other names:</b>	1,7-Octadiene, 3-methyl
<b>Inchi:</b>	InChI=1S/C9H16/c1-4-6-7-8-9(3)5-2/h4-5,9H,1-2,6-8H2,3H3
<b>InchiKey:</b>	LFXNEGVBADMEB-UHFFFAOYSA-N
<b>Formula:</b>	C9H16
<b>SMILES:</b>	C=CCCCC(C)C=C
<b>Mol. weight [g/mol]:</b>	124.22

## Physical Properties

Property code	Value	Unit	Source
gf	198.14	kJ/mol	Joback Method
hf	16.49	kJ/mol	Joback Method
hfus	12.98	kJ/mol	Joback Method
hvap	33.90	kJ/mol	Joback Method
log10ws	-3.06		Crippen Method
logp	3.165		Crippen Method
mcvol	129.070	ml/mol	McGowan Method
pc	2525.19	kPa	Joback Method
rinpol	821.00		NIST Webbook
rinpol	823.00		NIST Webbook
rinpol	823.00		NIST Webbook
rinpol	823.00		NIST Webbook
rinpol	830.00		NIST Webbook
rinpol	821.00		NIST Webbook
rinpol	821.00		NIST Webbook
tb	398.24	K	Joback Method
tc	571.80	K	Joback Method
tf	172.67	K	Joback Method
vc	0.495	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.06	J/mol×K	398.24	Joback Method
cpg	255.55	J/mol×K	427.17	Joback Method

cpg	268.46	J/mol×K	456.09	Joback Method
cpg	280.80	J/mol×K	485.02	Joback Method
cpg	292.60	J/mol×K	513.95	Joback Method
cpg	303.88	J/mol×K	542.87	Joback Method
cpg	314.64	J/mol×K	571.80	Joback Method
dvisc	0.0074402	Paxs	172.67	Joback Method
dvisc	0.0024687	Paxs	210.27	Joback Method
dvisc	0.0011447	Paxs	247.86	Joback Method
dvisc	0.0006499	Paxs	285.46	Joback Method
dvisc	0.0004209	Paxs	323.05	Joback Method
dvisc	0.0002985	Paxs	360.64	Joback Method
dvisc	0.0002258	Paxs	398.24	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R1942&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R1942&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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