

# Butanoic acid, 3-methyl-, 5-methyl-2-(1-methylethyl)cyclohexyl ester, (1«alpha»,2«beta»,5«alpha»)-3-Methylbutanoic acid (1«alpha»,2«beta»,5«alpha»)-3-Methylbutanoic acid Menthyl valerate

Validol

<b>Inchi:</b>	InChI=1S/C15H28O2/c1-5-6-7-15(16)17-14-10-12(4)8-9-13(14)11(2)3/h11-14H,5-10H2,1
<b>InchiKey:</b>	LCJPVSLESAPYMK-UHFFFAOYSA-N
<b>Formula:</b>	C15H28O2
<b>SMILES:</b>	CCCCC(=O)OC1CC(C)CCC1C(C)C
<b>Mol. weight [g/mol]:</b>	240.38
<b>CAS:</b>	89-47-4

## Physical Properties

Property code	Value	Unit	Source
gf	-151.91	kJ/mol	Joback Method
hf	-589.37	kJ/mol	Joback Method
hfus	27.85	kJ/mol	Joback Method
hvap	57.56	kJ/mol	Joback Method
log10ws	-4.25		Crippen Method
logp	4.181		Crippen Method
mcvol	218.790	ml/mol	McGowan Method
pc	1645.76	kPa	Joback Method
tb	628.66	K	Joback Method
tc	823.75	K	Joback Method
tf	314.87	K	Joback Method
vc	0.825	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	617.44	J/molxK	628.66	Joback Method
cpg	638.98	J/molxK	661.18	Joback Method
cpg	659.41	J/molxK	693.69	Joback Method
cpg	678.75	J/molxK	726.21	Joback Method
cpg	697.00	J/molxK	758.72	Joback Method
cpg	714.18	J/molxK	791.24	Joback Method

cpg	730.29	J/mol×K	823.75	Joback Method
dvisc	0.0030275	Paxs	314.87	Joback Method
dvisc	0.0013669	Paxs	367.17	Joback Method
dvisc	0.0007525	Paxs	419.47	Joback Method
dvisc	0.0004729	Paxs	471.76	Joback Method
dvisc	0.0003260	Paxs	524.06	Joback Method
dvisc	0.0002405	Paxs	576.36	Joback Method
dvisc	0.0001866	Paxs	628.66	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C89474&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C89474&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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