

# trans-1,2-Cyclohexanediamine

<b>Other names:</b>	trans-1,2-Cyclohexaneiamine 1,2-Cyclohexanediamine, trans- trans-1,2-Diaminocyclohexane 1,2-Cyclohexanediamine, (1R,2R)-rel- (+)-trans-1,2-Diaminocyclohexane
<b>Inchi:</b>	InChI=1S/C6H14N2/c7-5-3-1-2-4-6(5)8/h5-6H,1-4,7-8H2/t5-,6-/m0/s1
<b>InchiKey:</b>	SSJXIUAEKJCMH-WDSKDSINSA-N
<b>Formula:</b>	C6H14N2
<b>SMILES:</b>	NC1CCCCC1N
<b>Mol. weight [g/mol]:</b>	114.19
<b>CAS:</b>	1121-22-8

## Physical Properties

Property code	Value	Unit	Source
gf	149.28	kJ/mol	Joback Method
hf	-65.61	kJ/mol	Joback Method
hfus	14.60	kJ/mol	Joback Method
hvap	50.35	kJ/mol	Joback Method
log10ws	-1.32		Crippen Method
logp	0.215		Crippen Method
mcvol	104.500	ml/mol	McGowan Method
pc	4397.41	kPa	Joback Method
tb	461.00	K	NIST Webbook
tc	730.03	K	Joback Method
tf	327.04	K	Joback Method
vc	0.361	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	250.11	J/molxK	496.62	Joback Method
cpg	266.28	J/molxK	535.52	Joback Method
cpg	281.50	J/molxK	574.42	Joback Method
cpg	295.80	J/molxK	613.32	Joback Method

cpg	309.20	J/mol×K	652.22	Joback Method
cpg	321.72	J/mol×K	691.13	Joback Method
cpg	333.38	J/mol×K	730.03	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	353.20	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1121228&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1121228&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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