

# Isoeugenyl isovalerate

<b>Inchi:</b>	InChI=1S/C15H20O3/c1-5-6-12-7-8-13(14(10-12)17-4)18-15(16)9-11(2)3/h5-8,10-11H,9H
<b>InchiKey:</b>	JNVPTYXMIALTSY-AATRIKPKSA-N
<b>Formula:</b>	C15H20O3
<b>SMILES:</b>	CC=Cc1ccc(OC(=O)CC(C)C)c(OC)c1
<b>Mol. weight [g/mol]:</b>	248.32

## Physical Properties

Property code	Value	Unit	Source
gf	-92.57	kJ/mol	Joback Method
hf	-404.42	kJ/mol	Joback Method
hfus	28.52	kJ/mol	Joback Method
hvap	63.72	kJ/mol	Joback Method
log10ws	-4.16		Crippen Method
logp	3.680		Crippen Method
mcvol	207.460	ml/mol	McGowan Method
pc	1944.08	kPa	Joback Method
rinsol	1814.00		NIST Webbook
tb	681.67	K	Joback Method
tc	890.17	K	Joback Method
tf	384.58	K	Joback Method
vc	0.783	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	551.65	J/molxK	681.67	Joback Method
cpg	567.57	J/molxK	716.42	Joback Method
cpg	582.57	J/molxK	751.17	Joback Method
cpg	596.66	J/molxK	785.92	Joback Method
cpg	609.86	J/molxK	820.67	Joback Method
cpg	622.19	J/molxK	855.42	Joback Method
cpg	633.67	J/molxK	890.17	Joback Method
dvisc	0.0009687	Paxs	384.58	Joback Method
dvisc	0.0005108	Paxs	434.09	Joback Method

dvisc	0.0003070	Paxs	483.61	Joback Method
dvisc	0.0002028	Paxs	533.12	Joback Method
dvisc	0.0001438	Paxs	582.64	Joback Method
dvisc	0.0001076	Paxs	632.15	Joback Method
dvisc	0.0000840	Paxs	681.67	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R224945&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R224945&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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