

# 2,3,5,6-Tetrafluoroanisole

<b>Other names:</b>	2,3,5,6-Tetrafluoromethoxybenzene Benzene, 1,2,4,5-tetrafluoro-3-methoxy- Anisole, 2,3,5,6-tetrafluoro- 1,2,4,5-Tetrafluoro-3-methoxybenzene
<b>Inchi:</b>	InChI=1S/C7H4F4O/c1-12-7-5(10)3(8)2-4(9)6(7)11/h2H,1H3
<b>InchiKey:</b>	AXCOCGJDERQVDK-UHFFFAOYSA-N
<b>Formula:</b>	C7H4F4O
<b>SMILES:</b>	COc1c(F)c(F)cc(F)c1F
<b>Mol. weight [g/mol]:</b>	180.10
<b>CAS:</b>	2324-98-3

## Physical Properties

Property code	Value	Unit	Source
ea	0.22 ± 0.09	eV	NIST Webbook
gf	-802.29	kJ/mol	Joback Method
hf	-913.82	kJ/mol	Joback Method
hfus	19.88	kJ/mol	Joback Method
hvap	35.24	kJ/mol	Joback Method
log10ws	-2.92		Crippen Method
logp	2.252		Crippen Method
mcvol	98.680	ml/mol	McGowan Method
pc	3032.27	kPa	Joback Method
tb	411.20	K	NIST Webbook
tc	599.90	K	Joback Method
tf	269.74	K	Joback Method
vc	0.409	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	196.49	J/mol×K	425.66	Joback Method
cpg	203.76	J/mol×K	454.70	Joback Method
cpg	210.82	J/mol×K	483.74	Joback Method
cpg	217.67	J/mol×K	512.78	Joback Method

cpg	224.30	J/mol×K	541.82	Joback Method
cpg	230.72	J/mol×K	570.86	Joback Method
cpg	236.91	J/mol×K	599.90	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2324983&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2324983&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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