

# 1H-Indene, 1-methylene-

<b>Other names:</b>	1-Methylene-1H-indene 1-Methylene indene 1-Methylele-1H-indene
<b>Inchi:</b>	InChI=1S/C10H8/c1-8-6-7-9-4-2-3-5-10(8)9/h2-7H,1H2
<b>InchiKey:</b>	HVVZVBWIBBTXAJ-UHFFFAOYSA-N
<b>Formula:</b>	C10H8
<b>SMILES:</b>	C=C1C=Cc2ccccc21
<b>Mol. weight [g/mol]:</b>	128.17
<b>CAS:</b>	2471-84-3

## Physical Properties

Property code	Value	Unit	Source
gf	287.60	kJ/mol	Joback Method
hf	210.49	kJ/mol	Joback Method
hfus	12.44	kJ/mol	Joback Method
hvap	41.47	kJ/mol	Joback Method
log10ws	-3.01		Crippen Method
logp	2.727		Crippen Method
mcvol	108.540	ml/mol	McGowan Method
pc	3699.95	kPa	Joback Method
rinpol	193.80		NIST Webbook
rinpol	194.78		NIST Webbook
rinpol	193.40		NIST Webbook
rinpol	1098.00		NIST Webbook
rinpol	193.80		NIST Webbook
rinpol	1095.00		NIST Webbook
rinpol	1098.00		NIST Webbook
ripol	1763.00		NIST Webbook
ripol	1763.00		NIST Webbook
tb	469.59	K	Joback Method
tc	697.66	K	Joback Method
tf	278.02	K	Joback Method
vc	0.415	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	210.26	J/molxK	469.59	Joback Method
cpg	222.83	J/molxK	507.60	Joback Method
cpg	234.43	J/molxK	545.61	Joback Method
cpg	245.14	J/molxK	583.63	Joback Method
cpg	255.02	J/molxK	621.64	Joback Method
cpg	264.15	J/molxK	659.65	Joback Method
cpg	272.59	J/molxK	697.66	Joback Method
dvisc	0.0011345	Paxs	278.02	Joback Method
dvisc	0.0008696	Paxs	309.95	Joback Method
dvisc	0.0007004	Paxs	341.88	Joback Method
dvisc	0.0005854	Paxs	373.81	Joback Method
dvisc	0.0005033	Paxs	405.73	Joback Method
dvisc	0.0004424	Paxs	437.66	Joback Method
dvisc	0.0003957	Paxs	469.59	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2471843&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2471843&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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