

4,4,4-Trifluoro-1-bromobutane

Inchi:	InChI=1S/C4H6BrF3/c5-3-1-2-4(6,7)8/h1-3H2
InchiKey:	DBCAQXHNJOFNGC-UHFFFAOYSA-N
Formula:	C4H6BrF3
SMILES:	FC(F)(F)CCCB
Mol. weight [g/mol]:	190.99
CAS:	406-81-5

Physical Properties

Property code	Value	Unit	Source
gf	-584.47	kJ/mol	Joback Method
hf	-696.64	kJ/mol	Joback Method
hfus	13.23	kJ/mol	Joback Method
hvap	27.19	kJ/mol	Joback Method
log10ws	-2.59		Crippen Method
logp	2.724		Crippen Method
mvol	90.030	ml/mol	McGowan Method
pc	3677.55	kPa	Joback Method
tb	351.66	K	Joback Method
tc	518.63	K	Joback Method
tf	198.83	K	Joback Method
vc	0.364	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	153.13	J/mol×K	351.66	Joback Method
cpg	161.24	J/mol×K	379.49	Joback Method
cpg	168.88	J/mol×K	407.32	Joback Method
cpg	176.07	J/mol×K	435.15	Joback Method
cpg	182.83	J/mol×K	462.98	Joback Method
cpg	189.18	J/mol×K	490.81	Joback Method
cpg	195.14	J/mol×K	518.63	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C406815&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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