

Ethanamine, 1,1-dimethyl, N-(1-methylethyl)

Inchi:	InChI=1S/C7H17N/c1-6(2)8-7(3,4)5/h6,8H,1-5H3
InchiKey:	ZWXQPERWRDHCMZ-UHFFFAOYSA-N
Formula:	C7H17N
SMILES:	CC(C)NC(C)(C)C
Mol. weight [g/mol]:	115.22

Physical Properties

Property code	Value	Unit	Source
gf	97.85	kJ/mol	Joback Method
hf	-148.37	kJ/mol	Joback Method
hfus	8.05	kJ/mol	Joback Method
hvap	35.93	kJ/mol	Joback Method
log10ws	-2.17		Crippen Method
logp	1.783		Crippen Method
mcvol	119.470	ml/mol	McGowan Method
pc	2921.84	kPa	Joback Method
rinpol	696.00		NIST Webbook
tb	406.06	K	Joback Method
tc	590.51	K	Joback Method
tf	208.73	K	Joback Method
vc	0.446	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	239.77	J/mol×K	406.06	Joback Method
cpg	254.21	J/mol×K	436.80	Joback Method
cpg	267.92	J/mol×K	467.54	Joback Method
cpg	280.95	J/mol×K	498.29	Joback Method
cpg	293.32	J/mol×K	529.03	Joback Method
cpg	305.05	J/mol×K	559.77	Joback Method
cpg	316.17	J/mol×K	590.51	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R19751&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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