

# Diethylmalonic acid, 2,6-dimethoxyphenyl hexyl ester

<b>Inchi:</b>	InChI=1S/C21H32O6/c1-6-9-10-11-15-26-19(22)21(7-2,8-3)20(23)27-18-16(24-4)13-12-1
<b>InchiKey:</b>	SDKCPEGTXPMWRU-UHFFFAOYSA-N
<b>Formula:</b>	C21H32O6
<b>SMILES:</b>	CCCCCOC(=O)C(CC)(CC)C(=O)Oc1c(OC)cccc1OC
<b>Mol. weight [g/mol]:</b>	380.48

## Physical Properties

Property code	Value	Unit	Source
gf	-455.91	kJ/mol	Joback Method
hf	-1025.97	kJ/mol	Joback Method
hfus	43.94	kJ/mol	Joback Method
hvap	87.78	kJ/mol	Joback Method
log10ws	-5.25		Crippen Method
logp	4.539		Crippen Method
mvol	309.610	ml/mol	McGowan Method
pc	1225.98	kPa	Joback Method
rinpol	2489.00		NIST Webbook
rinpol	2489.00		NIST Webbook
tb	910.71	K	Joback Method
tc	1120.29	K	Joback Method
tf	569.09	K	Joback Method
vc	1.177	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	995.64	J/molxK	910.71	Joback Method
cpg	1010.80	J/molxK	945.64	Joback Method
cpg	1024.57	J/molxK	980.57	Joback Method
cpg	1036.97	J/molxK	1015.50	Joback Method
cpg	1048.01	J/molxK	1050.43	Joback Method
cpg	1057.71	J/molxK	1085.36	Joback Method
cpg	1066.07	J/molxK	1120.29	Joback Method
dvisc	0.0002013	Paxs	569.09	Joback Method

dvisc	0.0001162	Paxs	626.03	Joback Method
dvisc	0.0000735	Paxs	682.96	Joback Method
dvisc	0.0000499	Paxs	739.90	Joback Method
dvisc	0.0000358	Paxs	796.84	Joback Method
dvisc	0.0000268	Paxs	853.77	Joback Method
dvisc	0.0000208	Paxs	910.71	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369813&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369813&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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