

# Costunolide

<b>Other names:</b>	Germacra-1(10),4,11(13)-trien-12-oic acid, 6«alpha»-hydroxy-, «gamma»-lactone, (E,E)-Cyclodeca[b]furan-2(3H)-one, 3a,4,5,8,9,11a-hexahydro-6,10-dimethyl-3-methylene-, Costunolide, [3aS-(3aR*,6E,10E,11aS*)]-Costus lactone (E,E)-6-«alpha»-Hydroxygermacra-1(10),4,11(13)-trien-12-oic acid «gamma»-lactone 6,10-Dimethyl-3-methylene-3a,4,5,8,9,11a-hexahydrocyclodeca[b]furan-2(3H)-one-, (3aS,6E,10E,11aR)- N <sub>2</sub> C <sub>10</sub> 6704
<b>Inchi:</b>	InChI=1S/C15H20O2/c1-10-5-4-6-11(2)9-14-13(8-7-10)12(3)15(16)17-14/h5,9,13-14H,3-
<b>InchiKey:</b>	HRYLQFBHBWLLLL-GYIATTAWSA-N
<b>Formula:</b>	C15H20O2
<b>SMILES:</b>	<chem>C=C1C(=O)OC2C=C(C)CCC=C(C)CCC12</chem>
<b>Mol. weight [g/mol]:</b>	232.32
<b>CAS:</b>	553-21-9

## Physical Properties

Property code	Value	Unit	Source
gf	-2.75	kJ/mol	Joback Method
hf	-343.29	kJ/mol	Joback Method
hfus	24.17	kJ/mol	Joback Method
hvap	60.84	kJ/mol	Joback Method
log10ws	-4.18		Crippen Method
logp	3.551		Crippen Method
mcvol	195.030	ml/mol	McGowan Method
pc	2191.78	kPa	Joback Method
tb	688.18	K	Joback Method
tc	932.76	K	Joback Method
tf	405.08	K	Joback Method
vc	0.718	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	560.07	J/molxK	688.18	Joback Method
cpg	581.51	J/molxK	728.94	Joback Method

cpg	601.37	J/mol×K	769.71	Joback Method
cpg	619.64	J/mol×K	810.47	Joback Method
cpg	636.29	J/mol×K	851.23	Joback Method
cpg	651.32	J/mol×K	892.00	Joback Method
cpg	664.71	J/mol×K	932.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C553219&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C553219&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/64-028-0/Costunolide.pdf>

Generated by Cheméo on 2024-04-20 09:55:38.152628421 +0000 UTC m=+15896187.073205737.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.