

5,5,8,8-Tetramethylcyclodecene cis

Inchi:	InChI=1S/C14H26/c1-13(2)9-7-5-6-8-10-14(3,4)12-11-13/h5-6H,7-12H2,1-4H3/b6-5-
InchiKey:	GNLGQSTXGUKFPN-WAYWQWQTSA-N
Formula:	C14H26
SMILES:	CC1(C)CCC=CCCC(C)(C)CC1
Mol. weight [g/mol]:	194.36
CAS:	4277-41-2

Physical Properties

Property code	Value	Unit	Source
gf	54.32	kJ/mol	Joback Method
hf	-234.69	kJ/mol	Joback Method
hfus	5.15	kJ/mol	Joback Method
hvap	45.56	kJ/mol	Joback Method
log10ws	-4.95		Crippen Method
logp	4.949		Crippen Method
mcvol	192.960	ml/mol	McGowan Method
pc	2135.43	kPa	Joback Method
tb	551.32	K	Joback Method
tc	787.24	K	Joback Method
tf	285.16	K	Joback Method
vc	0.702	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	483.66	J/molxK	551.32	Joback Method
cpg	509.66	J/molxK	590.64	Joback Method
cpg	534.00	J/molxK	629.96	Joback Method
cpg	556.91	J/molxK	669.28	Joback Method
cpg	578.61	J/molxK	708.60	Joback Method
cpg	599.33	J/molxK	747.92	Joback Method
cpg	619.29	J/molxK	787.24	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C4277412&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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