

# 2,3,4,5,6-Pentabromobenzyl alcohol, 2-methylbutyl ether

Inchi:	InChI=1S/C12H13Br5O/c1-3-6(2)4-18-5-7-8(13)10(15)12(17)11(16)9(7)14/h6H,3-5H2,1-2
InchiKey:	SMQMBZTYGZGJEI-UHFFFAOYSA-N
Formula:	C12H13Br5O
SMILES:	CCC(C)COCc1c(Br)c(Br)c(Br)c(Br)c1Br
Mol. weight [g/mol]:	572.75

## Physical Properties

Property code	Value	Unit	Source
gf	78.58	kJ/mol	Joback Method
hf	-117.68	kJ/mol	Joback Method
hfus	43.02	kJ/mol	Joback Method
hvap	82.09	kJ/mol	Joback Method
log10ws	-9.10		Crippen Method
logp	7.062		Crippen Method
mcvol	249.550	ml/mol	McGowan Method
pc	3163.27	kPa	Joback Method
rinsol	2757.00		NIST Webbook
tb	878.32	K	Joback Method
tc	1137.99	K	Joback Method
tf	620.25	K	Joback Method
vc	0.921	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.01	J/molxK	878.32	Joback Method
cpg	586.61	J/molxK	1094.71	Joback Method
cpg	578.84	J/molxK	1051.43	Joback Method
cpg	570.67	J/molxK	1008.15	Joback Method
cpg	562.03	J/molxK	964.88	Joback Method
cpg	552.83	J/molxK	921.60	Joback Method
cpg	594.05	J/molxK	1137.99	Joback Method
dvisc	0.0000668	Paxs	878.32	Joback Method
dvisc	0.0000787	Paxs	835.31	Joback Method

dvisc	0.0000944	Paxs	792.30	Joback Method
dvisc	0.0001157	Paxs	749.28	Joback Method
dvisc	0.0001452	Paxs	706.27	Joback Method
dvisc	0.0001878	Paxs	663.26	Joback Method
dvisc	0.0002517	Paxs	620.25	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375313&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375313&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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