

# Tetrahydrofuran, 2-methyl-2-(4-methylpentyl)

Inchi:	InChI=1S/C11H22O/c1-10(2)6-4-7-11(3)8-5-9-12-11/h10H,4-9H2,1-3H3
InchiKey:	XCZUYIGFSVRKBU-UHFFFAOYSA-N
Formula:	C11H22O
SMILES:	CC(C)CCCC1(C)CCCO1
Mol. weight [g/mol]:	170.29

## Physical Properties

Property code	Value	Unit	Source
gf	-15.76	kJ/mol	Joback Method
hf	-331.93	kJ/mol	Joback Method
hfus	16.34	kJ/mol	Joback Method
hvap	43.31	kJ/mol	Joback Method
log10ws	-3.28		Crippen Method
logp	3.382		Crippen Method
mcvol	160.860	ml/mol	McGowan Method
pc	2379.54	kPa	Joback Method
ripol	1159.00		NIST Webbook
ripol	1325.00		NIST Webbook
ripol	1325.00		NIST Webbook
tb	493.11	K	Joback Method
tc	692.17	K	Joback Method
tf	260.10	K	Joback Method
vc	0.606	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.68	J/molxK	493.11	Joback Method
cpg	400.83	J/molxK	526.29	Joback Method
cpg	418.83	J/molxK	559.46	Joback Method
cpg	435.80	J/molxK	592.64	Joback Method
cpg	451.83	J/molxK	625.82	Joback Method
cpg	467.01	J/molxK	658.99	Joback Method
cpg	481.45	J/molxK	692.17	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R410250&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R410250&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ri<sub>npol</sub>:</b>	Non-polar retention indices
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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