

# Benzene, 1-(bromomethyl)-3-chloro-

<b>Other names:</b>	«alpha»-Bromo-m-chlorotoluene 1-(Bromomethyl)-3-chlorobenzene m-Chlorobenzyl bromide 3-Chlorobenzyl bromide Toluene, «alpha»-bromo-m-chloro- «alpha»-Bromo-3-chlorotoluene
<b>Inchi:</b>	InChI=1S/C7H6BrCl/c8-5-6-2-1-3-7(9)4-6/h1-4H,5H2
<b>InchiKey:</b>	LZIYAIRGDHSVED-UHFFFAOYSA-N
<b>Formula:</b>	C7H6BrCl
<b>SMILES:</b>	Clc1cccc(CBr)c1
<b>Mol. weight [g/mol]:</b>	205.48
<b>CAS:</b>	766-80-3

## Physical Properties

Property code	Value	Unit	Source
gf	113.23	kJ/mol	Joback Method
hf	47.84	kJ/mol	Joback Method
hfus	17.02	kJ/mol	Joback Method
hvap	44.93	kJ/mol	Joback Method
log10ws	-3.47		Crippen Method
logp	3.235		Crippen Method
mcvol	115.470	ml/mol	McGowan Method
pc	4205.63	kPa	Joback Method
tb	494.81	K	Joback Method
tc	734.55	K	Joback Method
tf	297.31	K	Joback Method
vc	0.430	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	195.64	J/mol×K	494.81	Joback Method
cpg	237.24	J/mol×K	694.60	Joback Method
cpg	230.19	J/mol×K	654.64	Joback Method

cpg	222.55	J/mol×K	614.68	Joback Method
cpg	214.28	J/mol×K	574.72	Joback Method
cpg	205.32	J/mol×K	534.77	Joback Method
cpg	243.74	J/mol×K	734.55	Joback Method
dvisc	0.0003139	Paxs	494.81	Joback Method
dvisc	0.0003844	Paxs	461.89	Joback Method
dvisc	0.0004857	Paxs	428.98	Joback Method
dvisc	0.0006381	Paxs	396.06	Joback Method
dvisc	0.0008806	Paxs	363.14	Joback Method
dvisc	0.0012960	Paxs	330.23	Joback Method
dvisc	0.0020778	Paxs	297.31	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	382.70	K	1.60	NIST Webbook
tbrp	382.20	K	1.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C766803&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C766803&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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