

# Acetamide, N-(2-methoxyphenyl)-

<b>Other names:</b>	o-Acetanisidide o-Methoxyacetanilide 2'-Methoxyacetanilide o-Acetanisidine N-Acetyl-O-anisidine Aceto-o-anisidine Acetanilide, 2'-methoxy- 2-Methoxyacetanilide Acetyl-O-anisidine 2-(Acetylamino)anisole NSC 4004 Anisidide, o-acet- N-(2-methoxyphenyl)acetamide
<b>Inchi:</b>	InChI=1S/C9H11NO2/c1-7(11)10-8-5-3-4-6-9(8)12-2/h3-6H,1-2H3,(H,10,11)
<b>InchiKey:</b>	FGOFNVXHDGQVVG-UHFFFAOYSA-N
<b>Formula:</b>	C9H11NO2
<b>SMILES:</b>	COc1ccccc1NC(C)=O
<b>Mol. weight [g/mol]:</b>	165.19
<b>CAS:</b>	93-26-5

## Physical Properties

Property code	Value	Unit	Source
chs	-4046.10 ± 1.50	kJ/mol	NIST Webbook
gf	-16.85	kJ/mol	Joback Method
hf	-195.36	kJ/mol	Joback Method
hfs	-1068.00	kJ/mol	NIST Webbook
hfus	20.60	kJ/mol	Joback Method
hvap	54.16	kJ/mol	Joback Method
log10ws	-1.79		Crippen Method
logp	1.654		Crippen Method
mcvol	131.330	ml/mol	McGowan Method
pc	3431.89	kPa	Joback Method
tb	577.20	K	NIST Webbook
tc	780.85	K	Joback Method
tf	354.95	K	Joback Method
vc	0.490	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.19	J/mol×K	563.44	Joback Method
cpg	313.64	J/mol×K	599.68	Joback Method
cpg	325.37	J/mol×K	635.91	Joback Method
cpg	336.41	J/mol×K	672.15	Joback Method
cpg	346.75	J/mol×K	708.38	Joback Method
cpg	356.42	J/mol×K	744.62	Joback Method
cpg	365.43	J/mol×K	780.85	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C93265&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C93265&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/64-177-5/Acetamide-N-2-methoxyphenyl.pdf>

Generated by Cheméo on 2024-04-26 06:39:54.686984285 +0000 UTC m=+16402843.607561596.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.