

# Malonic acid, 2,2-dichloroethyl ethyl ester

<b>Inchi:</b>	InChI=1S/C7H10Cl2O4/c1-2-12-6(10)3-7(11)13-4-5(8)9/h5H,2-4H2,1H3
<b>InchiKey:</b>	VTIZVXMFTSBZRS-UHFFFAOYSA-N
<b>Formula:</b>	C7H10Cl2O4
<b>SMILES:</b>	CCOC(=O)CC(=O)OCC(Cl)Cl
<b>Mol. weight [g/mol]:</b>	229.06

## Physical Properties

Property code	Value	Unit	Source
gf	-486.08	kJ/mol	Joback Method
hf	-714.17	kJ/mol	Joback Method
hfus	24.33	kJ/mol	Joback Method
hvap	57.87	kJ/mol	Joback Method
log10ws	-1.39		Crippen Method
logp	1.286		Crippen Method
mcvol	148.850	ml/mol	McGowan Method
pc	2878.12	kPa	Joback Method
rinpol	1367.00		NIST Webbook
rinpol	1367.00		NIST Webbook
tb	586.56	K	Joback Method
tc	785.49	K	Joback Method
tf	357.81	K	Joback Method
vc	0.568	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	326.69	J/molxK	586.56	Joback Method
cpg	336.43	J/molxK	619.72	Joback Method
cpg	345.69	J/molxK	652.87	Joback Method
cpg	354.48	J/molxK	686.03	Joback Method
cpg	362.79	J/molxK	719.18	Joback Method
cpg	370.60	J/molxK	752.34	Joback Method
cpg	377.92	J/molxK	785.49	Joback Method
dvisc	0.0021007	Paxs	357.81	Joback Method

dvisc	0.0011981	Paxs	395.94	Joback Method
dvisc	0.0007542	Paxs	434.06	Joback Method
dvisc	0.0005116	Paxs	472.18	Joback Method
dvisc	0.0003677	Paxs	510.31	Joback Method
dvisc	0.0002767	Paxs	548.43	Joback Method
dvisc	0.0002161	Paxs	586.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349058&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349058&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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