

# 2-Ethyl-1,2-dibromobutane

<b>Inchi:</b>	InChI=1S/C6H12Br2/c1-3-6(8,4-2)5-7/h3-5H2,1-2H3
<b>InchiKey:</b>	UBFOYEMXPUOMHZ-UHFFFAOYSA-N
<b>Formula:</b>	C6H12Br2
<b>SMILES:</b>	CCC(Br)(CC)CBr
<b>Mol. weight [g/mol]:</b>	243.97

## Physical Properties

Property code	Value	Unit	Source
gf	31.12	kJ/mol	Joback Method
hf	-123.26	kJ/mol	Joback Method
hfus	14.45	kJ/mol	Joback Method
hvap	40.52	kJ/mol	Joback Method
log10ws	-3.31		Crippen Method
logp	3.335		Crippen Method
mcvol	130.400	ml/mol	McGowan Method
pc	3731.66	kPa	Joback Method
rinpol	1124.00		NIST Webbook
rinpol	1124.00		NIST Webbook
tb	465.77	K	Joback Method
tc	680.86	K	Joback Method
tf	279.40	K	Joback Method
vc	0.484	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	237.14	J/molxK	465.77	Joback Method
cpg	248.60	J/molxK	501.62	Joback Method
cpg	259.25	J/molxK	537.47	Joback Method
cpg	269.14	J/molxK	573.31	Joback Method
cpg	278.32	J/molxK	609.16	Joback Method
cpg	286.87	J/molxK	645.01	Joback Method
cpg	294.82	J/molxK	680.86	Joback Method
dvisc	0.0041844	Paxs	279.40	Joback Method

dvisc	0.0022883	Paxs	310.46	Joback Method
dvisc	0.0013966	Paxs	341.52	Joback Method
dvisc	0.0009255	Paxs	372.58	Joback Method
dvisc	0.0006534	Paxs	403.65	Joback Method
dvisc	0.0004849	Paxs	434.71	Joback Method
dvisc	0.0003744	Paxs	465.77	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R559385&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R559385&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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