

# Benzyl 4-hydroxybenzoate

**Other names:**

Benzyl p-hydroxybenzoate  
p-Hydroxybenzoic acid benzyl ester  
Benzoic acid, 4-hydroxy-, phenylmethyl ester  
Benzoic acid, p-hydroxy-, benzyl ester  
Benzyl paraben  
Benzyl Parasept  
Benzyl Tegosept  
Nipabenzyl  
Parosept  
Solbrol Z  
NSC 8080

4-Hydroxybenzoic acid benzyl ester

**Inchi:**

InChI=1S/C14H12O3/c15-13-8-6-12(7-9-13)14(16)17-10-11-4-2-1-3-5-11/h1-9,15H,10H2

**InchiKey:**

MOZDKDIOPSPTBH-UHFFFAOYSA-N

**Formula:**

C14H12O3

**SMILES:**

O=C(OCc1cccc1)c1ccc(O)cc1

**Mol. weight [g/mol]:**

228.24

**CAS:**

94-18-8

## Physical Properties

Property code	Value	Unit	Source
gf	-96.72	kJ/mol	Joback Method
hf	-281.34	kJ/mol	Joback Method
hfus	28.67	kJ/mol	Joback Method
hvap	73.48	kJ/mol	Joback Method
log10ws	-3.37		Crippen Method
logp	2.749		Crippen Method
mcvol	173.910	ml/mol	McGowan Method
pc	3439.94	kPa	Joback Method
tb	729.99	K	Joback Method
tc	979.95	K	Joback Method
tf	484.26	K	Joback Method
vc	0.594	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	464.36	J/molxK	729.99	Joback Method
cpg	477.44	J/molxK	771.65	Joback Method
cpg	489.51	J/molxK	813.31	Joback Method
cpg	500.71	J/molxK	854.97	Joback Method
cpg	511.14	J/molxK	896.63	Joback Method
cpg	520.92	J/molxK	938.29	Joback Method
cpg	530.18	J/molxK	979.95	Joback Method
dvisc	0.0002882	Paxs	484.26	Joback Method
dvisc	0.0001379	Paxs	525.22	Joback Method
dvisc	0.0000734	Paxs	566.17	Joback Method
dvisc	0.0000426	Paxs	607.12	Joback Method
dvisc	0.0000264	Paxs	648.08	Joback Method
dvisc	0.0000174	Paxs	689.04	Joback Method
dvisc	0.0000120	Paxs	729.99	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C94188&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C94188&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/64-218-9/Benzyl-4-hydroxybenzoate.pdf>

Generated by Cheméo on 2024-04-18 13:46:35.789970837 +0000 UTC m=+15737244.710548152.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.