

# Dimethylmalonic acid, nonyl pentadecyl ester

<b>Inchi:</b>	InChI=1S/C29H56O4/c1-5-7-9-11-13-14-15-16-17-18-20-22-24-26-33-28(31)29(3,4)27(3)
<b>InchiKey:</b>	ITOHLHOOAJEMFW-UHFFFAOYSA-N
<b>Formula:</b>	C29H56O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCOC(=O)C(C)(C)C(=O)OCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	468.75

## Physical Properties

Property code	Value	Unit	Source
gf	-271.70	kJ/mol	Joback Method
hf	-1140.24	kJ/mol	Joback Method
hfus	69.03	kJ/mol	Joback Method
hvap	97.16	kJ/mol	Joback Method
log10ws	-9.45		Crippen Method
logp	8.941		Crippen Method
mvol	434.350	ml/mol	McGowan Method
pc	660.51	kPa	Joback Method
rinpol	3000.00		NIST Webbook
rinpol	3000.00		NIST Webbook
tb	1012.27	K	Joback Method
tc	1256.59	K	Joback Method
tf	563.33	K	Joback Method
vc	1.696	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1539.29	J/molxK	1012.27	Joback Method
cpg	1562.26	J/molxK	1052.99	Joback Method
cpg	1583.28	J/molxK	1093.71	Joback Method
cpg	1602.46	J/molxK	1134.43	Joback Method
cpg	1619.92	J/molxK	1175.15	Joback Method
cpg	1635.76	J/molxK	1215.87	Joback Method
cpg	1650.11	J/molxK	1256.59	Joback Method
dvisc	0.0002393	Paxs	563.33	Joback Method

dvisc	0.0001037	Paxs	638.15	Joback Method
dvisc	0.0000535	Paxs	712.98	Joback Method
dvisc	0.0000313	Paxs	787.80	Joback Method
dvisc	0.0000201	Paxs	862.62	Joback Method
dvisc	0.0000139	Paxs	937.45	Joback Method
dvisc	0.0000101	Paxs	1012.27	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361774&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361774&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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