

# 3-Methyl-4-cyclohexene-1,2-dicarboxylic anhydride

<b>Other names:</b>	1,3-Isobenzofurandione, 3a,4,7,7a-tetrahydro-4-methyl-Maleic anhydride and 1,3-pentadiene adduct 1,2,3,6-Tetrahydro-3-methylphthalic anhydride 3-Methyl-«delta»4-tetrahydrophthalic anhydride 3-Methyl-1,2,3,6-tetrahydrophthalic anhydride 3-Methyltetrahydrophthalic anhydride 4-Cyclohexene-1,2-dicarboxylic anhydride, 3-methyl-NSC 2352
<b>Inchi:</b>	InChI=1S/C9H10O3/c1-5-3-2-4-6-7(5)9(11)12-8(6)10/h2-3,5-7H,4H2,1H3
<b>InchiKey:</b>	XPEKVUUBSDFMDR-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O3
<b>SMILES:</b>	CC1C=CCC2C(=O)OC(=O)C12
<b>Mol. weight [g/mol]:</b>	166.17
<b>CAS:</b>	5333-84-6

## Physical Properties

Property code	Value	Unit	Source
gf	-198.95	kJ/mol	Joback Method
hf	-471.93	kJ/mol	Joback Method
hfus	18.33	kJ/mol	Joback Method
hvap	48.96	kJ/mol	Joback Method
log10ws	-1.15		Crippen Method
logp	0.898		Crippen Method
mcvol	120.660	ml/mol	McGowan Method
pc	3530.46	kPa	Joback Method
tb	588.69	K	Joback Method
tc	838.56	K	Joback Method
tf	376.04	K	Joback Method
vc	0.450	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.30	J/molxK	588.69	Joback Method

cpg	338.36	J/mol×K	630.34	Joback Method
cpg	354.38	J/mol×K	671.98	Joback Method
cpg	369.33	J/mol×K	713.63	Joback Method
cpg	383.21	J/mol×K	755.27	Joback Method
cpg	395.97	J/mol×K	796.92	Joback Method
cpg	407.61	J/mol×K	838.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5333846&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5333846&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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