

# Aminoacetonitrile

<b>Other names:</b>	H2NCH2CN Acetonitrile, amino- AAN «alpha»-Aminoacetonitrile 2-Aminoacetonitrile Cyanomethylamine Glycinenitrile Glycinonitrile NCCH2NH2
<b>Inchi:</b>	InChI=1S/C2H4N2/c3-1-2-4/h1,3H2
<b>InchiKey:</b>	DFNYGALUNNFWKJ-UHFFFAOYSA-N
<b>Formula:</b>	C2H4N2
<b>SMILES:</b>	N#CCN
<b>Mol. weight [g/mol]:</b>	56.07
<b>CAS:</b>	540-61-4

## Physical Properties

Property code	Value	Unit	Source
affp	824.90	kJ/mol	NIST Webbook
basg	791.00	kJ/mol	NIST Webbook
gf	165.59	kJ/mol	Joback Method
hf	114.06	kJ/mol	Joback Method
hfus	7.64	kJ/mol	Joback Method
hvap	41.17	kJ/mol	Joback Method
log10ws	0.04		Crippen Method
logp	-0.531		Crippen Method
mvol	50.400	ml/mol	McGowan Method
pc	5266.25	kPa	Joback Method
tb	419.77	K	Joback Method
tc	631.29	K	Joback Method
tf	260.55	K	Joback Method
vc	0.203	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	88.64	J/mol×K	419.77	Joback Method
cpg	92.71	J/mol×K	455.02	Joback Method
cpg	96.59	J/mol×K	490.28	Joback Method
cpg	100.29	J/mol×K	525.53	Joback Method
cpg	103.81	J/mol×K	560.78	Joback Method
cpg	107.17	J/mol×K	596.04	Joback Method
cpg	110.36	J/mol×K	631.29	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	331.20	K	2.00	NIST Webbook
tbrp	331.00	K	2.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C540614&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C540614&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>t<sub>brp</sub>:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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