

# Butanoic acid, propyl ester

<b>Other names:</b>	Butyric acid, propyl ester N-PROPYL N-BUTYRATE PROPYL BUTANOATE PROPYL BUTYRATE PROPYL ESTER BUTYRIC ACID Propyl ester of butanoic acid Propyl n-butyrate Propylester kyseliny maselne n-Butyric acid n-propyl ester n-Propanol butyrate n-Propyl butyrate propyl bytanoate
<b>Inchi:</b>	InChI=1S/C7H14O2/c1-3-5-7(8)9-6-4-2/h3-6H2,1-2H3
<b>InchiKey:</b>	HUAZGNHGCJGYNP-UHFFFAOYSA-N
<b>Formula:</b>	C7H14O2
<b>SMILES:</b>	CCCOC(=O)CCC
<b>Mol. weight [g/mol]:</b>	130.18
<b>CAS:</b>	105-66-8

## Physical Properties

Property code	Value	Unit	Source
chl	-4187.80	kJ/mol	NIST Webbook
dm	1.80	debye	KDB
gf	-225.86	kJ/mol	Joback Method
hf	-432.61	kJ/mol	Joback Method
hfus	16.67	kJ/mol	Joback Method
hvap	40.33	kJ/mol	Joback Method
log10ws	-1.92		Aqueous Solubility Prediction Method
logp	1.740		Crippen Method
mcvol	116.930	ml/mol	McGowan Method
pc	2710.00	kPa	KDB
pc	2720.00 ± 30.00	kPa	NIST Webbook
rinpol	135.60		NIST Webbook
rinpol	900.00		NIST Webbook
rinpol	891.00		NIST Webbook
rinpol	897.00		NIST Webbook

rinpol	881.50	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	885.00	NIST Webbook
rinpol	888.00	NIST Webbook
rinpol	880.00	NIST Webbook
rinpol	880.00	NIST Webbook
rinpol	875.00	NIST Webbook
rinpol	878.00	NIST Webbook
rinpol	884.00	NIST Webbook
rinpol	882.00	NIST Webbook
rinpol	884.00	NIST Webbook
rinpol	886.00	NIST Webbook
rinpol	879.00	NIST Webbook
rinpol	885.00	NIST Webbook
rinpol	885.00	NIST Webbook
rinpol	885.00	NIST Webbook
rinpol	885.00	NIST Webbook
rinpol	880.00	NIST Webbook
rinpol	892.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	875.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	916.00	NIST Webbook
rinpol	895.00	NIST Webbook
rinpol	865.00	NIST Webbook
rinpol	880.00	NIST Webbook
rinpol	895.00	NIST Webbook
rinpol	903.00	NIST Webbook
rinpol	897.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	898.00	NIST Webbook
rinpol	891.00	NIST Webbook
rinpol	898.00	NIST Webbook
rinpol	873.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	893.00	NIST Webbook
rinpol	879.00	NIST Webbook
rinpol	881.00	NIST Webbook
rinpol	880.00	NIST Webbook
rinpol	881.00	NIST Webbook
rinpol	135.60	NIST Webbook
rinpol	877.00	NIST Webbook
rinpol	881.00	NIST Webbook

rinpol	884.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	899.00	NIST Webbook
rinpol	875.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	881.00	NIST Webbook
rinpol	891.00	NIST Webbook
rinpol	875.00	NIST Webbook
rinpol	887.00	NIST Webbook
rinpol	896.00	NIST Webbook
rinpol	900.00	NIST Webbook
rinpol	881.00	NIST Webbook
rinpol	880.00	NIST Webbook
rinpol	906.00	NIST Webbook
rinpol	912.00	NIST Webbook
rinpol	882.00	NIST Webbook
rinpol	881.00	NIST Webbook
rinpol	876.00	NIST Webbook
ripol	1137.00	NIST Webbook
ripol	1107.00	NIST Webbook
ripol	1110.00	NIST Webbook
ripol	1137.00	NIST Webbook
ripol	1107.00	NIST Webbook
ripol	1133.00	NIST Webbook
ripol	1120.00	NIST Webbook
ripol	1135.00	NIST Webbook
ripol	1126.00	NIST Webbook
ripol	1153.00	NIST Webbook
ripol	1156.00	NIST Webbook
ripol	1117.00	NIST Webbook
ripol	1110.00	NIST Webbook
ripol	1123.00	NIST Webbook
ripol	1133.00	NIST Webbook
ripol	1116.00	NIST Webbook
ripol	1098.00	NIST Webbook
ripol	1122.00	NIST Webbook
ripol	1127.00	NIST Webbook
ripol	1123.00	NIST Webbook
ripol	1110.00	NIST Webbook
ripol	1110.00	NIST Webbook
ripol	1110.00	NIST Webbook
ripol	1110.00	NIST Webbook

ripol	1136.00		NIST Webbook
tb	416.20	K	KDB
tc	593.70 ± 0.60	K	NIST Webbook
tc	600.00	K	KDB
tc	599.80 ± 5.00	K	NIST Webbook
tc	592.50 ± 0.60	K	NIST Webbook
tf	177.90	K	KDB
tf	177.95	K	Aqueous Solubility Prediction Method
tf	178.00 ± 0.50	K	NIST Webbook
vc	0.452	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.48	J/mol×K	582.84	Joback Method
cpg	299.73	J/mol×K	612.24	Joback Method
cpg	238.65	J/mol×K	435.85	Joback Method
cpg	249.76	J/mol×K	465.25	Joback Method
cpg	260.50	J/mol×K	494.65	Joback Method
cpg	270.86	J/mol×K	524.04	Joback Method
cpg	280.85	J/mol×K	553.44	Joback Method
dvisc	0.0003401	Paxs	403.34	Joback Method
dvisc	0.0002656	Paxs	435.85	Joback Method
dvisc	0.0031909	Paxs	240.81	Joback Method
dvisc	0.0016480	Paxs	273.32	Joback Method
dvisc	0.0009795	Paxs	305.82	Joback Method
dvisc	0.0006434	Paxs	338.33	Joback Method
dvisc	0.0004549	Paxs	370.84	Joback Method
hvapt	42.00	kJ/mol	385.50	NIST Webbook
hvapt	44.30	kJ/mol	343.50	NIST Webbook
hvapt	39.60	kJ/mol	410.00	NIST Webbook
rfi	1.39500		303.15	Correlation and Prediction of Excess Quantities and Vapor-Liquid Equilibria of Alkyl Esters + tert-Butyl Alcohol: Experimental Data for Propyl Esters + tert-Butyl Alcohol

rho1	867.90	kg/m3	298.15	Thermodynamic study of (alkyl esters + a,x-alkyl dihalides) VII. HE m and VE m for 20 binary mixtures {xCu 1H2u 1CO2C3H7 + (1 x)a,x-ClCH2(CH2)v 2CH2Cl}, where u = 1 to 4, a = 1 and v = x = 2 to 6. An analysis of behavior using the COSMO-RS methodology
rho1	879.00	kg/m3	288.00	KDB

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	415.90	K	102.00	NIST Webbook
tbrp	312.40	K	1.90	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49207e+01
Coeff. B	-3.70105e+03
Coeff. C	-5.67560e+01
Temperature range (K), min.	309.68
Temperature range (K), max.	441.91

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	7.34865e+01
Coeff. B	-7.60502e+03
Coeff. C	-8.55915e+00

Coeff. D	5.25288e-06
Temperature range (K), min.	177.95
Temperature range (K), max.	594.00

## Sources

<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1088">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1088</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1088">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1088</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C105668&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C105668&amp;Units=SI</a>
<b>Correlation and Prediction of Excess Quantities and Vapor-Liquid Equilibria</b>	<a href="https://www.doi.org/10.1021/je060003c">https://www.doi.org/10.1021/je060003c</a>
<b>Grignani Method: tert-Butyl Alcohol:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Experimental Data for Propyl Esters + Aqueous Solubility Prediction Method: tert-Butyl Alcohol:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Thermodynamic study of (alkyl esters + a,x-alkyl dihalides) VII. HE m and VE m for 20 binary mixtures {xCu 1H2u 1CO2C3H7 + (1-x)a,x-CiCH2(CH2)v 2CH2Cl}, where u = 1 to 4, a = 1 and v = 1 to 4. Analysis of behavior using the COSMO-RS methodology:</b>	<a href="https://www.doi.org/10.1016/j.jct.2008.10.007">https://www.doi.org/10.1016/j.jct.2008.10.007</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dm:</b>	Dipole Moment
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rfi:</b>	Refractive Index
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tbrp:** Boiling point at reduced pressure  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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