

# Thiirane, 2,3-dimethyl-, trans-

<b>Other names:</b>	trans-2-Butene Episulfide Butane, 2,3-epithio-, trans- trans-2-Butene Sulfide trans-2,3-Dimethylthiirane (E)-2,3-Dimethylthiirane 2,3-Dimethyl-thiirane (E) trans-Dimethylthiirane
<b>Inchi:</b>	InChI=1S/C4H8S/c1-3-4(2)5-3/h3-4H,1-2H3/t3-,4-/m0/s1
<b>InchiKey:</b>	ZMJBHCIEMIVIFZ-IMJSIDKUSA-N
<b>Formula:</b>	C4H8S
<b>SMILES:</b>	CC1SC1C
<b>Mol. weight [g/mol]:</b>	88.17
<b>CAS:</b>	5955-98-6

## Physical Properties

Property code	Value	Unit	Source
chl	-3290.00 ± 1.00	kJ/mol	NIST Webbook
gf	75.70	kJ/mol	Joback Method
hf	-28.17	kJ/mol	Joback Method
hfl	-30.00 ± 1.00	kJ/mol	NIST Webbook
hfus	8.98	kJ/mol	Joback Method
hvap	29.91	kJ/mol	Joback Method
log10ws	-1.50		Crippen Method
logp	1.510		Crippen Method
mcvol	72.710	ml/mol	McGowan Method
pc	4379.97	kPa	Joback Method
rinpol	746.00		NIST Webbook
rinpol	688.00		NIST Webbook
tb	340.82	K	Joback Method
tc	537.89	K	Joback Method
tf	231.99	K	Joback Method
vc	0.262	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	115.68	J/mol×K	340.82	Joback Method
cpg	125.45	J/mol×K	373.67	Joback Method
cpg	134.70	J/mol×K	406.51	Joback Method
cpg	143.43	J/mol×K	439.36	Joback Method
cpg	151.68	J/mol×K	472.20	Joback Method
cpg	159.48	J/mol×K	505.05	Joback Method
cpg	166.84	J/mol×K	537.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5955986&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5955986&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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