

# Benzenemethanol, 4-methoxy-, formate

<b>Other names:</b>	Benzyl alcohol, p-methoxy-, formate p-Methoxybenzyl alcohol, formate Anisyl alcohol, formate Anisyl formate 4-Methoxybenzyl formate p-Anisyl formate p-Methoxybenzyl formate
<b>Inchi:</b>	InChI=1S/C9H10O3/c1-11-9-4-2-8(3-5-9)6-12-7-10/h2-5,7H,6H2,1H3
<b>InchiKey:</b>	XPDORSROGAZEGY-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O3
<b>SMILES:</b>	<chem>COc1ccc(COC=O)cc1</chem>
<b>Mol. weight [g/mol]:</b>	166.17
<b>CAS:</b>	122-91-8

## Physical Properties

Property code	Value	Unit	Source
gf	-181.84	kJ/mol	Joback Method
hf	-354.05	kJ/mol	Joback Method
hfus	17.38	kJ/mol	Joback Method
hvap	50.11	kJ/mol	Joback Method
log10ws	-1.75		Crippen Method
logp	1.368		Crippen Method
mcvol	127.220	ml/mol	McGowan Method
pc	3356.75	kPa	Joback Method
rinpol	1300.00		NIST Webbook
rinpol	1300.00		NIST Webbook
rinpol	1321.00		NIST Webbook
rinpol	1327.00		NIST Webbook
rinpol	1307.00		NIST Webbook
rinpol	1332.00		NIST Webbook
rinpol	1327.00		NIST Webbook
rinpol	1321.00		NIST Webbook
tb	530.48	K	Joback Method
tc	740.30	K	Joback Method
tf	316.59	K	Joback Method
vc	0.484	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	280.27	J/molxK	530.48	Joback Method
cpg	333.19	J/molxK	705.33	Joback Method
cpg	323.73	J/molxK	670.36	Joback Method
cpg	313.70	J/molxK	635.39	Joback Method
cpg	303.12	J/molxK	600.42	Joback Method
cpg	291.97	J/molxK	565.45	Joback Method
cpg	342.08	J/molxK	740.30	Joback Method
dvisc	0.0002028	Paxs	530.48	Joback Method
dvisc	0.0002512	Paxs	494.83	Joback Method
dvisc	0.0003217	Paxs	459.18	Joback Method
dvisc	0.0004295	Paxs	423.54	Joback Method
dvisc	0.0006047	Paxs	387.89	Joback Method
dvisc	0.0009125	Paxs	352.24	Joback Method
dvisc	0.0015106	Paxs	316.59	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C122918&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C122918&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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