

# Dichlorophen, O-(6-bromohexanoyl)-

<b>Inchi:</b>	InChI=1S/C19H19BrCl2O3/c20-9-3-1-2-4-19(24)25-18-8-6-16(22)12-14(18)10-13-11-15(
<b>InchiKey:</b>	DIHIWNGGTNTBLI-UHFFFAOYSA-N
<b>Formula:</b>	C19H19BrCl2O3
<b>SMILES:</b>	O=C(CCCCCBr)Oc1ccc(Cl)cc1Cc1cc(Cl)ccc1O
<b>Mol. weight [g/mol]:</b>	446.16

## Physical Properties

Property code	Value	Unit	Source
gf	-93.05	kJ/mol	Joback Method
hf	-424.10	kJ/mol	Joback Method
hfus	54.13	kJ/mol	Joback Method
hvap	101.80	kJ/mol	Joback Method
log10ws	-6.91		Crippen Method
logp	6.151		Crippen Method
mvol	286.340	ml/mol	McGowan Method
pc	2001.91	kPa	Joback Method
rinpol	3557.00		NIST Webbook
rinpol	3557.00		NIST Webbook
tb	1000.35	K	Joback Method
tc	1247.06	K	Joback Method
tf	697.81	K	Joback Method
vc	1.034	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	809.04	J/molxK	1000.35	Joback Method
cpg	821.33	J/molxK	1041.47	Joback Method
cpg	833.17	J/molxK	1082.59	Joback Method
cpg	844.68	J/molxK	1123.71	Joback Method
cpg	856.01	J/molxK	1164.82	Joback Method
cpg	867.28	J/molxK	1205.94	Joback Method
cpg	878.61	J/molxK	1247.06	Joback Method
dvisc	0.0000196	Paxs	697.81	Joback Method

dvisc	0.0000115	Paxs	748.23	Joback Method
dvisc	0.0000072	Paxs	798.66	Joback Method
dvisc	0.0000048	Paxs	849.08	Joback Method
dvisc	0.0000033	Paxs	899.50	Joback Method
dvisc	0.0000024	Paxs	949.93	Joback Method
dvisc	0.0000018	Paxs	1000.35	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354723&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354723&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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