

# Pentaethylene glycol, decyl ether

<b>Other names:</b>	2-(2-(2-(2-(2-decyloxy-ethoxy)-ethoxy)-ethoxy)-ethoxy)-ethanol
<b>Inchi:</b>	InChI=1S/C20H42O6/c1-2-3-4-5-6-7-8-9-11-22-13-15-24-17-19-26-20-18-25-16-14-23-12
<b>InchiKey:</b>	QAXPOSPGRHYIHE-UHFFFAOYSA-N
<b>Formula:</b>	C20H42O6
<b>SMILES:</b>	CCCCCCCCCOCCOCCOCCOCCOCCO
<b>Mol. weight [g/mol]:</b>	378.54

## Physical Properties

Property code	Value	Unit	Source
gf	-544.30	kJ/mol	Joback Method
hf	-1269.46	kJ/mol	Joback Method
hfus	57.58	kJ/mol	Joback Method
hvap	88.84	kJ/mol	Joback Method
log10ws	-2.89		Crippen Method
logp	3.202		Crippen Method
mcvol	327.880	ml/mol	McGowan Method
pc	1020.08	kPa	Joback Method
rinpol	2733.70		NIST Webbook
rinpol	2733.70		NIST Webbook
tb	861.28	K	Joback Method
tc	1057.53	K	Joback Method
tf	487.13	K	Joback Method
vc	1.264	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1090.17	J/molxK	861.28	Joback Method
cpg	1109.41	J/molxK	893.99	Joback Method
cpg	1127.28	J/molxK	926.70	Joback Method
cpg	1143.77	J/molxK	959.41	Joback Method
cpg	1158.88	J/molxK	992.11	Joback Method
cpg	1172.60	J/molxK	1024.82	Joback Method
cpg	1184.92	J/molxK	1057.53	Joback Method

dvisc	0.0002105	Paxs	487.13	Joback Method
dvisc	0.0000734	Paxs	549.49	Joback Method
dvisc	0.0000317	Paxs	611.85	Joback Method
dvisc	0.0000160	Paxs	674.20	Joback Method
dvisc	0.0000091	Paxs	736.56	Joback Method
dvisc	0.0000056	Paxs	798.92	Joback Method
dvisc	0.0000037	Paxs	861.28	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R184124&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R184124&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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