

# Ethanol, 2,2'-(butylimino)bis-

<b>Other names:</b>	Ethanol, 2,2'-(butylimino)di- Butylbis(2-hydroxyethyl)amine Butyldiethanolamine N-Butyl-N,N-bis(hydroxyethyl)amine N-Butyl-N,N-bis(2-hydroxyethyl)amine N-Butyl-2,2'-iminodiethanol N-Butyldiethanolamine N,N-Bis(2-hydroxyethyl)butylamine 2,2'-(Butylimino)diethanol N-n-Butyldiethanolamine 2,2'-(n-Butylimino)diethanol BIDE Bis(«beta»-hydroxyethyl)butylamine 2-(N-Butyl-N-2-hydroxyethylamino)ethanol NSC 60214
<b>Inchi:</b>	InChI=1S/C8H19NO2/c1-2-3-4-9(5-7-10)6-8-11/h10-11H,2-8H2,1H3
<b>InchiKey:</b>	GVNHOISKXMSMPX-UHFFFAOYSA-N
<b>Formula:</b>	C8H19NO2
<b>SMILES:</b>	CCCCN(CCO)CCO
<b>Mol. weight [g/mol]:</b>	161.24
<b>CAS:</b>	102-79-4

## Physical Properties

Property code	Value	Unit	Source
gf	-146.38	kJ/mol	Joback Method
hf	-445.38	kJ/mol	Joback Method
hfus	27.67	kJ/mol	Joback Method
hvap	68.80	kJ/mol	Joback Method
log10ws	-0.27		Crippen Method
logp	0.073		Crippen Method
mcvol	145.300	ml/mol	McGowan Method
pc	3072.75	kPa	Joback Method
tb	579.24	K	Joback Method
tc	735.48	K	Joback Method
tf	334.03	K	Joback Method
vc	0.539	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	380.29	J/mol×K	579.24	Joback Method
cpg	391.00	J/mol×K	605.28	Joback Method
cpg	401.27	J/mol×K	631.32	Joback Method
cpg	411.11	J/mol×K	657.36	Joback Method
cpg	420.53	J/mol×K	683.40	Joback Method
cpg	429.55	J/mol×K	709.44	Joback Method
cpg	438.18	J/mol×K	735.48	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	547.20	K	98.80	NIST Webbook

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C102794&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C102794&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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