

# 3,4-Diazanona-3,5-diene

Inchi:	InChI=1S/C7H14N2/c1-3-5-6-7-9-8-4-2/h6-7H,3-5H2,1-2H3/b7-6+,9-8+
InchiKey:	QXUWPEMSORWKIV-BLHCBFLLSA-N
Formula:	C7H14N2
SMILES:	CCCC=CN=NCC
Mol. weight [g/mol]:	126.20
CAS:	82946-55-2

## Physical Properties

Property code	Value	Unit	Source
hf	-23.37	kJ/mol	Joback Method
hvap	37.80	kJ/mol	Joback Method
ie	8.10	eV	NIST Webbook
log10ws	-2.36		Crippen Method
logp	2.772		Crippen Method
mcvol	120.850	ml/mol	McGowan Method
pc	2267.57	kPa	Joback Method
tb	512.92	K	Joback Method
tc	718.48	K	Joback Method

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C82946552&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C82946552&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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