

# Isopropyl phenyl ketone

<b>Other names:</b>	1-Propanone, 2-methyl-1-phenyl- Isobutyrophenone «alpha»-Methylpropiophenone Phenyl isopropyl ketone 2-Methylpropiophenone 2-Methyl-1-phenyl-1-propanone
<b>Inchi:</b>	InChI=1S/C10H12O/c1-8(2)10(11)9-6-4-3-5-7-9/h3-8H,1-2H3
<b>InchiKey:</b>	BSMGLVDZZMBWQB-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O
<b>SMILES:</b>	CC(C)C(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	148.20
<b>CAS:</b>	611-70-1

## Physical Properties

Property code	Value	Unit	Source
basg	839.30	kJ/mol	NIST Webbook
gf	14.37	kJ/mol	Joback Method
hf	-131.06	kJ/mol	Joback Method
hfus	13.77	kJ/mol	Joback Method
hvap	46.49	kJ/mol	Joback Method
log10ws	-2.73		Crippen Method
logp	2.525		Crippen Method
mcvol	129.570	ml/mol	McGowan Method
pc	3170.40	kPa	Joback Method
rinpol	1185.00		NIST Webbook
rinpol	1174.00		NIST Webbook
rinpol	1185.00		NIST Webbook
tb	490.20	K	NIST Webbook
tc	728.18	K	Joback Method
tf	263.81	K	Joback Method
vc	0.487	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.87	J/mol×K	508.31	Joback Method
cpg	339.99	J/mol×K	691.53	Joback Method
cpg	329.01	J/mol×K	654.89	Joback Method
cpg	317.24	J/mol×K	618.24	Joback Method
cpg	304.66	J/mol×K	581.60	Joback Method
cpg	291.21	J/mol×K	544.95	Joback Method
cpg	350.22	J/mol×K	728.18	Joback Method
dvisc	0.0002449	Paxs	508.31	Joback Method
dvisc	0.0003205	Paxs	467.56	Joback Method
dvisc	0.0004416	Paxs	426.81	Joback Method
dvisc	0.0006510	Paxs	386.06	Joback Method
dvisc	0.0010518	Paxs	345.31	Joback Method
dvisc	0.0019321	Paxs	304.56	Joback Method
dvisc	0.0042828	Paxs	263.81	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	493.20	K	99.50	NIST Webbook
tbrp	359.20	K	0.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C611701&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C611701&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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