

# Chlorphenesin

<b>Other names:</b>	1,2-Propanediol, 3-(4-chlorophenoxy)- 1,2-Propanediol, 3-(p-chlorophenoxy)- p-Chlorophenyl glyceryl ether p-Chlorophenyl-«alpha»-glyceryl ether Adermykon Chlorophenesin Demykon Gecophen Glycerol «alpha»-p-chlorophenyl ether Mycil 3-(p-Chlorophenoxy)-1,2-propanediol 3-(p-Chlorophenoxy)propane-1,2-diol 3-(4-Chlorophenoxy)-1,2-propanediol Gechophen NSC 6401
<b>Inchi:</b>	InChI=1S/C9H11ClO3/c10-7-1-3-9(4-2-7)13-6-8(12)5-11/h1-4,8,11-12H,5-6H2
<b>InchiKey:</b>	MXOAEAUPQDYUQM-UHFFFAOYSA-N
<b>Formula:</b>	C9H11ClO3
<b>SMILES:</b>	OCC(O)COc1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	202.63
<b>CAS:</b>	104-29-0

## Physical Properties

Property code	Value	Unit	Source
gf	-265.33	kJ/mol	Joback Method
hf	-461.73	kJ/mol	Joback Method
hfus	22.76	kJ/mol	Joback Method
hvap	78.33	kJ/mol	Joback Method
log10ws	-1.75		Crippen Method
logp	1.072		Crippen Method
mvol	143.760	ml/mol	McGowan Method
pc	3773.04	kPa	Joback Method
tb	680.75	K	Joback Method
tc	872.56	K	Joback Method
tf	388.92	K	Joback Method
vc	0.530	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	359.81	J/molxK	680.75	Joback Method
cpg	368.68	J/molxK	712.72	Joback Method
cpg	377.04	J/molxK	744.69	Joback Method
cpg	384.88	J/molxK	776.66	Joback Method
cpg	392.24	J/molxK	808.63	Joback Method
cpg	399.11	J/molxK	840.59	Joback Method
cpg	405.52	J/molxK	872.56	Joback Method
dvisc	0.0031082	Paxs	388.92	Joback Method
dvisc	0.0007638	Paxs	437.56	Joback Method
dvisc	0.0002486	Paxs	486.20	Joback Method
dvisc	0.0000992	Paxs	534.84	Joback Method
dvisc	0.0000461	Paxs	583.47	Joback Method
dvisc	0.0000241	Paxs	632.11	Joback Method
dvisc	0.0000139	Paxs	680.75	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C104290&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C104290&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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