

# Diclofop methyl

**Other names:**

(.+/-)-Diclofop-methyl  
2-(4-(2,4-Dichlorophenoxy)phenoxy)-methyl-propionate  
2-(4-(2,4-Dichlorophenoxy)phenoxy)propanoic acid methyl ester  
Dichlorfop-methyl  
Dicloflop-methyl  
Diclofop, methyl ester  
HOE 23408  
Hoegrass  
Hoelon  
Hoelon 3EC  
Illoxan  
Illoxan  
Methyl 2-(4-(2,4-dichlorophenoxy)phenoxy)propionate  
Methyl diclofop  
Propanoic acid, 2-[4-(2,4-dichlorophenoxy)phenoxy]-, methyl ester  
Propionic acid, 2-(4-(2,4-dichlorophenoxy)phenoxy)-, methyl ester  
diclofop-methyl

**Inchi:** InChI=1S/C16H14Cl2O4/c1-10(16(19)20-2)21-12-4-6-13(7-5-12)22-15-8-3-11(17)9-14(15)**InchiKey:** BACHBFVBHLGWSL-UHFFFAOYSA-N**Formula:** C16H14Cl2O3**SMILES:** COC(=O)C(C)Oc1ccc(Oc2ccc(Cl)cc2Cl)cc1**Mol. weight [g/mol]:** 325.19**CAS:** 51338-27-3

## Physical Properties

Property code	Value	Unit	Source
gf	-190.45	kJ/mol	Joback Method
hf	-480.92	kJ/mol	Joback Method
hfus	34.14	kJ/mol	Joback Method
hvap	80.11	kJ/mol	Joback Method
log10ws	-3.82		Aqueous Solubility Prediction Method
logp	4.726		Crippen Method
mcvol	232.440	ml/mol	McGowan Method
pc	2083.12	kPa	Joback Method
rinpol	2395.00		NIST Webbook
rinpol	2339.00		NIST Webbook

rmpol	2395.00		NIST Webbook
rmpol	2339.00		NIST Webbook
tb	829.33	K	Joback Method
tc	1067.40	K	Joback Method
tf	315.26 ± 0.20	K	NIST Webbook
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	616.47	J/mol×K	829.33	Joback Method
cpg	628.73	J/mol×K	869.01	Joback Method
cpg	639.72	J/mol×K	908.69	Joback Method
cpg	649.44	J/mol×K	948.36	Joback Method
cpg	657.91	J/mol×K	988.04	Joback Method
cpg	665.11	J/mol×K	1027.72	Joback Method
cpg	671.04	J/mol×K	1067.40	Joback Method
dvisc	0.0003732	Paxs	521.94	Joback Method
dvisc	0.0002341	Paxs	573.17	Joback Method
dvisc	0.0001585	Paxs	624.40	Joback Method
dvisc	0.0001139	Paxs	675.63	Joback Method
dvisc	0.0000857	Paxs	726.87	Joback Method
dvisc	0.0000670	Paxs	778.10	Joback Method
dvisc	0.0000540	Paxs	829.33	Joback Method
hfust	27.08	kJ/mol	314.40	NIST Webbook

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C51338273&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/64-542-9/Diclofop-methyl.pdf>

Generated by Cheméo on 2024-04-27 15:14:00.838059527 +0000 UTC m=+16520089.758636840.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.