

# Tetradecanethiol

<b>Other names:</b>	1-Mercaptotetradecane 1-Tetradecanethiol Myristyl mercaptan Tetradecanethiol-(1) Tetradecyl mercaptan n-Tetradecyl mercaptan tetradecane-1-thiol
<b>Inchi:</b>	InChI=1S/C14H30S/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15/h15H,2-14H2,1H3
<b>InchiKey:</b>	GEKDEMCKPCKTKEC-UHFFFAOYSA-N
<b>Formula:</b>	C14H30S
<b>SMILES:</b>	CCCCCCCCCCCCCS
<b>Mol. weight [g/mol]:</b>	230.45
<b>CAS:</b>	2079-95-0

## Physical Properties

Property code	Value	Unit	Source
gf	96.39	kJ/mol	Joback Method
hf	-293.81	kJ/mol	Joback Method
hfus	36.06	kJ/mol	Joback Method
hvap	53.50	kJ/mol	Joback Method
log10ws	-5.75		Crippen Method
logp	5.617		Crippen Method
mvol	224.470	ml/mol	McGowan Method
pc	1602.56	kPa	Joback Method
tb	582.58	K	Joback Method
tc	757.99	K	Joback Method
tf	284.00	K	Joback Method
vc	0.874	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	655.56	J/mol×K	728.76	Joback Method
cpg	571.84	J/mol×K	582.58	Joback Method

cpg	590.12	J/mol×K	611.82	Joback Method
cpg	607.61	J/mol×K	641.05	Joback Method
cpg	624.33	J/mol×K	670.29	Joback Method
cpg	640.31	J/mol×K	699.52	Joback Method
cpg	670.11	J/mol×K	757.99	Joback Method
cpl	501.57	J/mol×K	300.00	NIST Webbook
hvapt	67.30	kJ/mol	530.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.58515e+01
Coeff. B	-5.31782e+03
Coeff. C	-1.02212e+02
Temperature range (K), min.	443.89
Temperature range (K), max.	606.75

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2079950&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2079950&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>h<sub>vapt</sub>:</b>	Enthalpy of vaporization at a given temperature
<b>log<sub>10</sub>ws:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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