

«alpha»-Elemenone

Inchi:	InChI=1S/C15H22O/c1-7-15(6)9-14(16)12(10(2)3)8-13(15)11(4)5/h7-8,11H,1,9H2,2-6H3
InchiKey:	UYBFSLEXEYFGANT-OAHLLOKOSA-N
Formula:	C15H22O
SMILES:	<chem>C=CC1(C)CC(=O)C(=C(C)C)C=C1C(C)C</chem>
Mol. weight [g/mol]:	218.33

Physical Properties

Property code	Value	Unit	Source
gf	114.43	kJ/mol	Joback Method
hf	-188.37	kJ/mol	Joback Method
hfus	14.70	kJ/mol	Joback Method
hvap	53.27	kJ/mol	Joback Method
log10ws	-4.35		Crippen Method
logp	4.070		Crippen Method
mcvol	200.020	ml/mol	McGowan Method
pc	1944.08	kPa	Joback Method
rinpol	1670.00		NIST Webbook
rinpol	1670.00		NIST Webbook
rinpol	1670.00		NIST Webbook
tb	637.11	K	Joback Method
tc	863.01	K	Joback Method
tf	351.23	K	Joback Method
vc	0.758	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	529.15	J/molxK	637.11	Joback Method
cpg	548.65	J/molxK	674.76	Joback Method
cpg	567.15	J/molxK	712.41	Joback Method
cpg	584.79	J/molxK	750.06	Joback Method
cpg	601.66	J/molxK	787.71	Joback Method
cpg	617.90	J/molxK	825.36	Joback Method
cpg	633.61	J/molxK	863.01	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R324491&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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