

# 3-hydroxy-2-methyltetrahydrofuran

Inchi:	InChI=1S/C5H10O2/c1-4-5(6)2-3-7-4/h4-6H,2-3H2,1H3
InchiKey:	ZFVWBUVYNPLIIS-UHFFFAOYSA-N
Formula:	C5H10O2
SMILES:	CC1OCCC1O
Mol. weight [g/mol]:	102.13

## Physical Properties

Property code	Value	Unit	Source
gf	-202.88	kJ/mol	Joback Method
hf	-390.62	kJ/mol	Joback Method
hfus	15.78	kJ/mol	Joback Method
hvap	47.86	kJ/mol	Joback Method
log10ws	-0.38		Crippen Method
logp	0.156		Crippen Method
mcvol	82.190	ml/mol	McGowan Method
pc	4640.32	kPa	Joback Method
rinpol	816.00		NIST Webbook
rinpol	816.00		NIST Webbook
tb	443.54	K	Joback Method
tc	632.60	K	Joback Method
tf	240.16	K	Joback Method
vc	0.295	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	178.25	J/molxK	443.54	Joback Method
cpg	189.10	J/molxK	475.05	Joback Method
cpg	199.45	J/molxK	506.56	Joback Method
cpg	209.31	J/molxK	538.07	Joback Method
cpg	218.70	J/molxK	569.58	Joback Method
cpg	227.62	J/molxK	601.09	Joback Method
cpg	236.10	J/molxK	632.60	Joback Method
dvisc	0.0319611	Paxs	240.16	Joback Method

dvisc	0.0093294	Paxs	274.06	Joback Method
dvisc	0.0035712	Paxs	307.95	Joback Method
dvisc	0.0016538	Paxs	341.85	Joback Method
dvisc	0.0008800	Paxs	375.75	Joback Method
dvisc	0.0005198	Paxs	409.64	Joback Method
dvisc	0.0003327	Paxs	443.54	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R332695&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R332695&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/64-568-1/3-hydroxy-2-methyltetrahydrofuran.pdf>

Generated by Cheméo on 2025-12-25 04:57:16.343386352 +0000 UTC m=+6386833.873427005.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.