

# Cyclohexane, 1-decyl-2-methyl, cis

<b>Inchi:</b>	InChI=1S/C17H34/c1-3-4-5-6-7-8-9-10-14-17-15-12-11-13-16(17)2/h16-17H,3-15H2,1-2H
<b>InchiKey:</b>	VVIPGXSGFRNNAW-DLBZAZTESA-N
<b>Formula:</b>	C17H34
<b>SMILES:</b>	CCCCCCCCC1CCCC1C
<b>Mol. weight [g/mol]:</b>	238.45

## Physical Properties

Property code	Value	Unit	Source
gf	109.00	kJ/mol	Joback Method
hf	-360.23	kJ/mol	Joback Method
hfus	32.69	kJ/mol	Joback Method
hvap	53.56	kJ/mol	Joback Method
log10ws	-6.35		Crippen Method
logp	6.344		Crippen Method
mcvol	239.530	ml/mol	McGowan Method
pc	1394.37	kPa	Joback Method
rinsol	1713.00		NIST Webbook
tb	603.24	K	Joback Method
tc	784.96	K	Joback Method
tf	284.49	K	Joback Method
vc	0.919	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	662.67	J/molxK	603.24	Joback Method
cpg	685.88	J/molxK	633.53	Joback Method
cpg	707.99	J/molxK	663.81	Joback Method
cpg	729.03	J/molxK	694.10	Joback Method
cpg	749.04	J/molxK	724.39	Joback Method
cpg	768.03	J/molxK	754.68	Joback Method
cpg	786.03	J/molxK	784.96	Joback Method
dvisc	0.0046878	Paxs	284.49	Joback Method
dvisc	0.0017440	Paxs	337.62	Joback Method

dvisc	0.0008489	Paxs	390.74	Joback Method
dvisc	0.0004910	Paxs	443.87	Joback Method
dvisc	0.0003192	Paxs	496.99	Joback Method
dvisc	0.0002255	Paxs	550.12	Joback Method
dvisc	0.0001694	Paxs	603.24	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R553738&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R553738&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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