

# Disulfide, 1-propenyl propyl, trans

<b>Other names:</b>	(E)-1-propenyl propyl disulfide (E)-propenyl propyl disulfide propyl (E)-propenyl disulfide Propyl 1-propenyl disulphide, trans (E)-propenyl propyl disulphide (E) n-Propyl 1-propenyl disulfide Propyl 1-propenyl disulfide, trans Disulfide, (E)-1-propenyl, propyl propyl trans-1-propenyl disulfide
<b>Inchi:</b>	InChI=1S/C6H12S2/c1-3-5-7-8-6-4-2/h3,5H,4,6H2,1-2H3/b5-3+
<b>InchiKey:</b>	AAPBYIVJOWCMGH-HWKANZROSA-N
<b>Formula:</b>	C6H12S2
<b>SMILES:</b>	CC=CSSCCC
<b>Mol. weight [g/mol]:</b>	148.29

## Physical Properties

Property code	Value	Unit	Source
gf	146.10	kJ/mol	Joback Method
hf	33.79	kJ/mol	Joback Method
hfus	19.76	kJ/mol	Joback Method
hvap	42.54	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.311		Crippen Method
mcvol	123.800	ml/mol	McGowan Method
pc	3427.87	kPa	Joback Method
ripol	1096.00		NIST Webbook
ripol	1090.60		NIST Webbook
ripol	1099.00		NIST Webbook
ripol	1098.00		NIST Webbook
ripol	1098.00		NIST Webbook
ripol	1098.00		NIST Webbook
ripol	1100.00		NIST Webbook
ripol	1447.00		NIST Webbook
ripol	1422.00		NIST Webbook
ripol	1447.00		NIST Webbook
ripol	1473.00		NIST Webbook
ripol	1435.00		NIST Webbook

ripol	1438.00		NIST Webbook
ripol	1482.00		NIST Webbook
tb	478.40	K	Joback Method
tc	701.27	K	Joback Method
tf	221.10	K	Joback Method
vc	0.460	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	236.88	J/mol×K	478.40	Joback Method
cpg	248.67	J/mol×K	515.55	Joback Method
cpg	259.83	J/mol×K	552.69	Joback Method
cpg	270.40	J/mol×K	589.84	Joback Method
cpg	280.37	J/mol×K	626.98	Joback Method
cpg	289.77	J/mol×K	664.13	Joback Method
cpg	298.61	J/mol×K	701.27	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R88948&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R88948&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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