

# 4-Hydroperoxy-n-heptane

<b>Inchi:</b>	InChI=1S/C7H16O2/c1-3-5-7(9-8)6-4-2/h7-8H,3-6H2,1-2H3
<b>InchiKey:</b>	MRKPKEMCNSQLLX-UHFFFAOYSA-N
<b>Formula:</b>	C7H16O2
<b>SMILES:</b>	CCCC(CCC)OO
<b>Mol. weight [g/mol]:</b>	132.20
<b>CAS:</b>	761-40-0

## Physical Properties

Property code	Value	Unit	Source
chl	-4707.00 ± 2.00	kJ/mol	NIST Webbook
gf	-236.20	kJ/mol	Joback Method
hf	-477.54	kJ/mol	Joback Method
hfus	15.64	kJ/mol	Joback Method
hvap	49.88	kJ/mol	Joback Method
log10ws	-2.27		Crippen Method
logp	2.445		Crippen Method
mcvol	121.230	ml/mol	McGowan Method
pc	3062.55	kPa	Joback Method
tb	473.72	K	Joback Method
tc	637.45	K	Joback Method
tf	236.70	K	Joback Method
vc	0.459	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	271.52	J/mol×K	473.72	Joback Method
cpg	321.47	J/mol×K	610.16	Joback Method
cpg	312.18	J/mol×K	582.87	Joback Method
cpg	302.54	J/mol×K	555.59	Joback Method
cpg	292.56	J/mol×K	528.30	Joback Method
cpg	282.22	J/mol×K	501.01	Joback Method
cpg	330.41	J/mol×K	637.45	Joback Method
dvisc	0.0001517	Paxs	473.72	Joback Method

dvisc	0.0002622	Paxs	434.22	Joback Method
dvisc	0.0005057	Paxs	394.71	Joback Method
dvisc	0.0011285	Paxs	355.21	Joback Method
dvisc	0.0030786	Paxs	315.71	Joback Method
dvisc	0.0111917	Paxs	276.20	Joback Method
dvisc	0.0625947	Paxs	236.70	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C761400&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C761400&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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