

# 3,5-Octadiyne

<b>Other names:</b>	Diethyldiacetylene
<b>Inchi:</b>	InChI=1S/C8H10/c1-3-5-7-8-6-4-2/h3-4H2,1-2H3
<b>InchiKey:</b>	LILZEAJBVQOINI-UHFFFAOYSA-N
<b>Formula:</b>	C8H10
<b>SMILES:</b>	CCC#CC#CCC
<b>Mol. weight [g/mol]:</b>	106.17
<b>CAS:</b>	16387-70-5

## Physical Properties

Property code	Value	Unit	Source
gf	422.08	kJ/mol	Joback Method
hf	336.15	kJ/mol	Joback Method
hfus	22.72	kJ/mol	Joback Method
hvap	37.71	kJ/mol	Joback Method
ie	8.78	eV	NIST Webbook
ie	8.82	eV	NIST Webbook
ie	8.55	eV	NIST Webbook
log10ws	-2.76		Crippen Method
logp	1.813		Crippen Method
mvol	106.380	ml/mol	McGowan Method
pc	3602.88	kPa	Joback Method
rinpol	972.00		NIST Webbook
tb	437.00 ± 4.00	K	NIST Webbook
tc	616.12	K	Joback Method
tf	392.12	K	Joback Method
vc	0.407	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	182.91	J/mol×K	400.44	Joback Method
cpg	193.56	J/mol×K	436.39	Joback Method
cpg	203.76	J/mol×K	472.33	Joback Method
cpg	213.53	J/mol×K	508.28	Joback Method

cpg	222.88	J/mol×K	544.23	Joback Method
cpg	231.82	J/mol×K	580.18	Joback Method
cpg	240.36	J/mol×K	616.12	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.62366e+01
Coeff. B	-4.35058e+03
Coeff. C	-6.25390e+01
Temperature range (K), min.	335.32
Temperature range (K), max.	460.76

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16387705&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16387705&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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