

# Isopropyl stearate

<b>Other names:</b>	1-Methylethyl octadecanoate Octadecanoic acid, 1-methylethyl ester Octadecanoic acid, isopropyl ester Revenge Stearic acid, isopropyl ester Tegosoft S Wickenol 127
<b>Inchi:</b>	InChI=1S/C21H42O2/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-21(22)23-20(2)3/h2
<b>InchiKey:</b>	ZPWFUIUNWDIYCJ-UHFFFAOYSA-N
<b>Formula:</b>	C21H42O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCC(=O)OC(C)C
<b>Mol. weight [g/mol]:</b>	326.56
<b>CAS:</b>	112-10-7

## Physical Properties

Property code	Value	Unit	Source
gf	-110.42	kJ/mol	Joback Method
hf	-726.85	kJ/mol	Joback Method
hfus	49.41	kJ/mol	Joback Method
hvap	71.11	kJ/mol	Joback Method
log10ws	-7.59		Crippen Method
logp	7.200		Crippen Method
mvol	314.190	ml/mol	McGowan Method
pc	988.88	kPa	Joback Method
rinpol	2217.00		NIST Webbook
tb	755.73	K	Joback Method
tc	930.28	K	Joback Method
tf	383.59	K	Joback Method
vc	1.230	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1076.91	J/mol×K	930.28	Joback Method

cpg	969.75	J/mol×K	755.73	Joback Method
cpg	989.99	J/mol×K	784.82	Joback Method
cpg	1009.24	J/mol×K	813.91	Joback Method
cpg	1027.53	J/mol×K	843.00	Joback Method
cpg	1044.89	J/mol×K	872.09	Joback Method
cpg	1061.34	J/mol×K	901.19	Joback Method
dvisc	0.0000590	Paxs	755.73	Joback Method
dvisc	0.0018724	Paxs	383.59	Joback Method
dvisc	0.0007045	Paxs	445.61	Joback Method
dvisc	0.0003366	Paxs	507.64	Joback Method
dvisc	0.0001889	Paxs	569.66	Joback Method
dvisc	0.0001187	Paxs	631.68	Joback Method
dvisc	0.0000811	Paxs	693.71	Joback Method
hvapt	76.60	kJ/mol	468.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.93960e+01
Coeff. B	-7.53425e+03
Coeff. C	-1.31510e+02
Temperature range (K), min.	525.80
Temperature range (K), max.	666.44

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C112107&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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