

# Fumaric acid, heptyl 3-oxobut-2-yl ester

<b>Inchi:</b>	InChI=1S/C15H24O5/c1-4-5-6-7-8-11-19-14(17)9-10-15(18)20-13(3)12(2)16/h9-10,13H,4
<b>InchiKey:</b>	LZRBSRFCMYHGGG-MDZDMXLPSA-N
<b>Formula:</b>	C15H24O5
<b>SMILES:</b>	CCCCCCCOC(=O)C=CC(=O)OC(C)C(C)=O
<b>Mol. weight [g/mol]:</b>	284.35

## Physical Properties

Property code	Value	Unit	Source
gf	-443.56	kJ/mol	Joback Method
hf	-843.17	kJ/mol	Joback Method
hfus	38.46	kJ/mol	Joback Method
hvap	73.61	kJ/mol	Joback Method
log10ws	-3.07		Crippen Method
logp	2.577		Crippen Method
mcvol	234.360	ml/mol	McGowan Method
pc	1675.53	kPa	Joback Method
rinqol	1962.00		NIST Webbook
tb	752.77	K	Joback Method
tc	943.28	K	Joback Method
tf	432.98	K	Joback Method
vc	0.903	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	675.61	J/molxK	752.77	Joback Method
cpg	739.97	J/molxK	911.53	Joback Method
cpg	728.73	J/molxK	879.77	Joback Method
cpg	716.69	J/molxK	848.02	Joback Method
cpg	703.83	J/molxK	816.27	Joback Method
cpg	690.14	J/molxK	784.52	Joback Method
cpg	750.43	J/molxK	943.28	Joback Method
dvisc	0.0000820	Paxs	752.77	Joback Method
dvisc	0.0001080	Paxs	699.47	Joback Method

dvisc	0.0001488	Paxs	646.17	Joback Method
dvisc	0.0002172	Paxs	592.88	Joback Method
dvisc	0.0003417	Paxs	539.58	Joback Method
dvisc	0.0005938	Paxs	486.28	Joback Method
dvisc	0.0011820	Paxs	432.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348821&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348821&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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