

# Meclofenoxate

**Other names:**

Acetic acid, (4-chlorophenoxy)-, 2-(dimethylamino)ethyl ester  
(p-Chlorophenoxy)acetic acid «beta»-(dimethylamino)ethyl ester  
(p-Chlorophenoxy)acetic acid 2-(dimethylamino)ethyl ester  
2-(Dimethylamino)ethyl p-chlorophenoxyacetate  
2-(Dimethylamino)ethyl-4-(chlorophenoxy)acetate  
Acephene  
Acetic acid, (p-chlorophenoxy)-, 2-(dimethylamino)ethyl ester  
Analux  
ANP 235  
Cerebon  
Cetrexin  
Clofenoxin  
Clophenoxate  
Deanol p-chlorophenoxyacetate  
Deanolestere  
EN 1627  
Lacidril  
Meclofenoxane  
Meclophenoxate  
Mucidril  
Proseryl  
2-(Dimethylamino)ethyl (p-chlorophenoxy)acetate  
Centrexin  
Clopenoxin  
Acephen  
Clocete  
NSC 169411

**Inchi:**

InChI=1S/C12H16ClNO3/c1-14(2)7-8-16-12(15)9-17-11-5-3-10(13)4-6-11/h3-6H,7-9H2,1

**InchiKey:**

XZTYGFHClAKPGJ-UHFFFAOYSA-N

**Formula:**

C12H16ClNO3

**SMILES:**

CN(C)CCOC(=O)COc1ccc(Cl)cc1

**Mol. weight [g/mol]:**

257.71

**CAS:**

51-68-3

## Physical Properties

Property code	Value	Unit	Source
---------------	-------	------	--------

gf	-87.13		kJ/mol	Joback Method
hf	-391.18		kJ/mol	Joback Method
hfus	31.68		kJ/mol	Joback Method
hvap	63.24		kJ/mol	Joback Method
log10ws	-1.80			Crippen Method
logp	1.824			Crippen Method
mcvol	191.710		ml/mol	McGowan Method
pc	2347.36		kPa	Joback Method
rinpol	1790.00			NIST Webbook
rinpol	1770.00			NIST Webbook
rinpol	1790.00			NIST Webbook
tb	654.20		K	Joback Method
tc	860.18		K	Joback Method
tf	420.72		K	Joback Method
vc	0.709		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	484.21	J/mol×K	654.20	Joback Method
cpg	498.32	J/mol×K	688.53	Joback Method
cpg	511.57	J/mol×K	722.86	Joback Method
cpg	523.99	J/mol×K	757.19	Joback Method
cpg	535.59	J/mol×K	791.52	Joback Method
cpg	546.38	J/mol×K	825.85	Joback Method
cpg	556.37	J/mol×K	860.18	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C51683&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/64-815-6/Meclofenoxate.pdf>

Generated by Cheméo on 2024-04-25 19:47:53.364181973 +0000 UTC m=+16363722.284759289.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.