

atisirene

Inchi:	InChI=1S/C21H36/c1-18(2)9-6-10-20(5)16(18)8-12-21-11-7-15(13-17(20)21)19(3,4)14-2
InchiKey:	OFOGPDDZNWXWEO-PWLWMKPSSA-N
Formula:	C21H36
SMILES:	CC1(C)CC23CCC1CC2C1(C)CCCC(C)(C)C1CC3
Mol. weight [g/mol]:	288.51

Physical Properties

Property code	Value	Unit	Source
gf	275.45	kJ/mol	Joback Method
hf	-210.27	kJ/mol	Joback Method
hfus	12.31	kJ/mol	Joback Method
hvap	57.15	kJ/mol	Joback Method
log10ws	-6.50		Crippen Method
logp	6.445		Crippen Method
mvol	263.310	ml/mol	McGowan Method
pc	1558.58	kPa	Joback Method
rinpol	2054.00		NIST Webbook
tb	710.87	K	Joback Method
tc	954.62	K	Joback Method
tf	466.99	K	Joback Method
vc	0.997	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	850.56	J/mol×K	710.87	Joback Method
cpg	880.31	J/mol×K	751.50	Joback Method
cpg	909.67	J/mol×K	792.12	Joback Method
cpg	939.30	J/mol×K	832.75	Joback Method
cpg	969.84	J/mol×K	873.37	Joback Method
cpg	1001.96	J/mol×K	914.00	Joback Method
cpg	1036.31	J/mol×K	954.62	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R221293&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
r inpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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