

# Glutaric acid, 2-methyl-4-chlorophenyl pentyl ester

|                      |                                                                                  |
|----------------------|----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C17H23ClO4/c1-3-4-5-11-21-16(19)7-6-8-17(20)22-15-10-9-14(18)12-13(15)2 |
| InchiKey:            | VPWIFGCCNHKJPZ-UHFFFAOYSA-N                                                      |
| Formula:             | C17H23ClO4                                                                       |
| SMILES:              | CCCCCOC(=O)CCCC(=O)Oc1ccc(Cl)cc1C                                                |
| Mol. weight [g/mol]: | 326.81                                                                           |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -294.36 | kJ/mol  | Joback Method  |
| hf            | -685.96 | kJ/mol  | Joback Method  |
| hfus          | 42.82   | kJ/mol  | Joback Method  |
| hvap          | 79.73   | kJ/mol  | Joback Method  |
| log10ws       | -5.16   |         | Crippen Method |
| logp          | 4.458   |         | Crippen Method |
| mcvol         | 253.750 | ml/mol  | McGowan Method |
| pc            | 1607.71 | kPa     | Joback Method  |
| rinpol        | 2390.00 |         | NIST Webbook   |
| rinpol        | 2390.00 |         | NIST Webbook   |
| tb            | 815.01  | K       | Joback Method  |
| tc            | 1020.37 | K       | Joback Method  |
| tf            | 507.05  | K       | Joback Method  |
| vc            | 0.977   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 730.36    | J/mol×K | 815.01          | Joback Method |
| cpg           | 744.56    | J/mol×K | 849.24          | Joback Method |
| cpg           | 757.73    | J/mol×K | 883.46          | Joback Method |
| cpg           | 769.90    | J/mol×K | 917.69          | Joback Method |
| cpg           | 781.06    | J/mol×K | 951.91          | Joback Method |
| cpg           | 791.23    | J/mol×K | 986.14          | Joback Method |
| cpg           | 800.43    | J/mol×K | 1020.37         | Joback Method |
| dvisc         | 0.0005664 | Pa×s    | 507.05          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0003441 | Pa×s | 558.38 | Joback Method |
| dvisc | 0.0002273 | Pa×s | 609.70 | Joback Method |
| dvisc | 0.0001601 | Pa×s | 661.03 | Joback Method |
| dvisc | 0.0001187 | Pa×s | 712.36 | Joback Method |
| dvisc | 0.0000915 | Pa×s | 763.68 | Joback Method |
| dvisc | 0.0000730 | Pa×s | 815.01 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359015&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359015&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

Latest version available from:

<https://www.chemeo.com/cid/65-001-8/Glutaric-acid-2-methyl-4-chlorophenyl-pentyl-ester.pdf>

Generated by Cheméo on 2025-12-24 00:19:49.221284901 +0000 UTC m=+6283786.751325557.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.