

# Di-n-octadecyl sebacate

<b>Inchi:</b>	InChI=1S/C46H90O4/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-31-35-39-43-49-45(47)41-
<b>InchiKey:</b>	NTUPDNAURMTJID-UHFFFAOYSA-N
<b>Formula:</b>	C46H90O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)CCCCCCCC(=O)OCCCCCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	707.20
<b>CAS:</b>	3072-03-5

## Physical Properties

Property code	Value	Unit	Source
gf	-131.40	kJ/mol	Joback Method
hf	-1482.37	kJ/mol	Joback Method
hfus	120.47	kJ/mol	Joback Method
hvap	197.70 ± 6.90	kJ/mol	NIST Webbook
log10ws	-16.80		Crippen Method
logp	15.717		Crippen Method
mcvol	673.880	ml/mol	McGowan Method
pc	325.59	kPa	Joback Method
tb	1404.46	K	Joback Method
tc	2283.25	K	Joback Method
tf	752.50	K	Joback Method
vc	2.659	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2667.84	J/molxK	1404.46	Joback Method
cpg	2713.41	J/molxK	1550.92	Joback Method
cpg	2745.32	J/molxK	1697.39	Joback Method
cpg	2771.27	J/molxK	1843.85	Joback Method
cpg	2798.95	J/molxK	1990.32	Joback Method
cpg	2836.08	J/molxK	2136.78	Joback Method
cpg	2890.35	J/molxK	2283.25	Joback Method
cpl	1481.00	J/molxK	353.00	NIST Webbook
dvisc	0.0000212	Paxs	752.50	Joback Method

dvisc	0.0000086	Paxs	861.16	Joback Method
dvisc	0.0000043	Paxs	969.82	Joback Method
dvisc	0.0000024	Paxs	1078.48	Joback Method
dvisc	0.0000015	Paxs	1187.14	Joback Method
dvisc	0.0000011	Paxs	1295.80	Joback Method
dvisc	0.0000008	Paxs	1404.46	Joback Method
hvapt	157.50	kJ/mol	480.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3072035&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3072035&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>cp<sub>l</sub>:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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