

# Diethyl perfluoroglutarate

**Other names:**

Pentanedioic acid, hexafluoro-, diethyl ester  
Diethyl hexafluoroglutarate  
Hexafluoroglutaric acid diethyl ester  
Glutaric acid, hexafluoro-, diethyl ester  
Diethylester kyseliny hexafluorglutarove  
Hexafluoroglutaric acid diethyl ester  
Hexafluoropentanedioic acid diethyl ester  
Diethyl 2,2,3,3,4,4-hexafluoropentanedioate  
NSC 63360  
Pentanedioic acid, 2,2,3,3,4,4-hexafluoro-, 1,5-diethyl ester

**Inchi:**

InChI=1S/C9H10F6O4/c1-3-18-5(16)7(10,11)9(14,15)8(12,13)6(17)19-4-2/h3-4H2,1-2H3

**InchiKey:**

MSDPXVBLFJODJO-UHFFFAOYSA-N

**Formula:**

C9H10F6O4

**SMILES:**

CCOC(=O)C(F)(F)C(F)(F)C(F)(F)C(=O)OCC

**Mol. weight [g/mol]:**

296.16

**CAS:**

424-40-8

## Physical Properties

Property code	Value	Unit	Source
gf	-1603.28	kJ/mol	Joback Method
hf	-1921.60	kJ/mol	Joback Method
hfus	20.88	kJ/mol	Joback Method
hvap	45.15	kJ/mol	Joback Method
log10ws	-2.26		Crippen Method
logp	2.019		Crippen Method
mcvol	163.170	ml/mol	McGowan Method
pc	2060.49	kPa	Joback Method
tb	543.83	K	Joback Method
tc	707.82	K	Joback Method
tf	346.31	K	Joback Method
vc	0.662	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	427.58	J/mol×K	543.83	Joback Method
cpg	438.86	J/mol×K	571.16	Joback Method
cpg	449.48	J/mol×K	598.49	Joback Method
cpg	459.46	J/mol×K	625.82	Joback Method
cpg	468.82	J/mol×K	653.16	Joback Method
cpg	477.59	J/mol×K	680.49	Joback Method
cpg	485.80	J/mol×K	707.82	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	350.50 ± 2.50	K	0.40	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C424408&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C424408&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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