

# Oxirane, (3,3-dimethylbutyl)-

<b>Inchi:</b>	InChI=1S/C8H16O/c1-8(2,3)5-4-7-6-9-7/h7H,4-6H2,1-3H3
<b>InchiKey:</b>	WAHCIOYRXXNHEZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H16O
<b>SMILES:</b>	CC(C)(C)CCC1CO1
<b>Mol. weight [g/mol]:</b>	128.21
<b>CAS:</b>	53907-77-0

## Physical Properties

Property code	Value	Unit	Source
gf	-6.05	kJ/mol	Joback Method
hf	-276.40	kJ/mol	Joback Method
hfus	15.18	kJ/mol	Joback Method
hvap	36.53	kJ/mol	Joback Method
log10ws	-2.02		Crippen Method
logp	2.212		Crippen Method
mcvol	118.590	ml/mol	McGowan Method
pc	2934.52	kPa	Joback Method
tb	412.90	K	Joback Method
tc	603.19	K	Joback Method
tf	226.85	K	Joback Method
vc	0.451	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	249.38	J/molxK	412.90	Joback Method
cpg	265.13	J/molxK	444.61	Joback Method
cpg	279.95	J/molxK	476.33	Joback Method
cpg	293.90	J/molxK	508.04	Joback Method
cpg	307.00	J/molxK	539.76	Joback Method
cpg	319.33	J/molxK	571.47	Joback Method
cpg	330.91	J/molxK	603.19	Joback Method
dvisc	0.0036221	Paxs	226.85	Joback Method
dvisc	0.0021146	Paxs	257.86	Joback Method

dvisc	0.0013857	Paxs	288.87	Joback Method
dvisc	0.0009856	Paxs	319.88	Joback Method
dvisc	0.0007445	Paxs	350.88	Joback Method
dvisc	0.0005886	Paxs	381.89	Joback Method
dvisc	0.0004821	Paxs	412.90	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C53907770&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C53907770&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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