

# Phthalic acid, 3-ethylphenyl isoheptyl ester

<b>Inchi:</b>	InChI=1S/C22H26O4/c1-4-17-10-7-11-18(15-17)26-22(24)20-13-6-5-12-19(20)21(23)25-
<b>InchiKey:</b>	AKVQNGODRHWOKS-UHFFFAOYSA-N
<b>Formula:</b>	C22H26O4
<b>SMILES:</b>	CCc1cccc(OC(=O)c2ccccc2C(=O)OCCCC(C)C)c1
<b>Mol. weight [g/mol]:</b>	354.44

## Physical Properties

Property code	Value	Unit	Source
gf	-130.36	kJ/mol	Joback Method
hf	-542.17	kJ/mol	Joback Method
hfus	42.09	kJ/mol	Joback Method
hvap	88.37	kJ/mol	Joback Method
log10ws	-6.46		Crippen Method
logp	5.061		Crippen Method
mvol	288.200	ml/mol	McGowan Method
pc	1481.57	kPa	Joback Method
rinpol	2586.00		NIST Webbook
tb	918.22	K	Joback Method
tc	1142.92	K	Joback Method
tf	544.90	K	Joback Method
vc	1.093	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	897.27	J/molxK	918.22	Joback Method
cpg	911.59	J/molxK	955.67	Joback Method
cpg	924.54	J/molxK	993.12	Joback Method
cpg	936.15	J/molxK	1030.57	Joback Method
cpg	946.47	J/molxK	1068.02	Joback Method
cpg	955.52	J/molxK	1105.47	Joback Method
cpg	963.35	J/molxK	1142.92	Joback Method
dvisc	0.0004055	Paxs	544.90	Joback Method
dvisc	0.0002255	Paxs	607.12	Joback Method

dvisc	0.0001398	Paxs	669.34	Joback Method
dvisc	0.0000941	Paxs	731.56	Joback Method
dvisc	0.0000673	Paxs	793.78	Joback Method
dvisc	0.0000506	Paxs	856.00	Joback Method
dvisc	0.0000395	Paxs	918.22	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357080&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357080&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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