

# Chloroacetic acid, 4-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C8H6Cl2O2/c9-5-8(11)12-7-3-1-6(10)2-4-7/h1-4H,5H2
<b>InchiKey:</b>	GEQGYTKAVRDKNR-UHFFFAOYSA-N
<b>Formula:</b>	C8H6Cl2O2
<b>SMILES:</b>	O=C(CCl)Oc1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	205.04

## Physical Properties

Property code	Value	Unit	Source
gf	-138.52	kJ/mol	Joback Method
hf	-259.67	kJ/mol	Joback Method
hfus	21.31	kJ/mol	Joback Method
hvap	54.27	kJ/mol	Joback Method
log10ws	-2.62		Crippen Method
logp	2.484		Crippen Method
mvol	131.740	ml/mol	McGowan Method
pc	3468.36	kPa	Joback Method
rinpol	1424.00		NIST Webbook
tb	565.25	K	Joback Method
tc	795.82	K	Joback Method
tf	350.86	K	Joback Method
vc	0.497	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	259.71	J/molxK	565.25	Joback Method
cpg	301.95	J/molxK	757.40	Joback Method
cpg	294.73	J/molxK	718.97	Joback Method
cpg	286.90	J/molxK	680.54	Joback Method
cpg	278.47	J/molxK	642.11	Joback Method
cpg	269.41	J/molxK	603.68	Joback Method
cpg	308.59	J/molxK	795.82	Joback Method
dvisc	0.0002441	Paxs	565.25	Joback Method
dvisc	0.0003011	Paxs	529.52	Joback Method

dvisc	0.0003828	Paxs	493.79	Joback Method
dvisc	0.0005053	Paxs	458.06	Joback Method
dvisc	0.0006989	Paxs	422.32	Joback Method
dvisc	0.0010266	Paxs	386.59	Joback Method
dvisc	0.0016308	Paxs	350.86	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307712&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307712&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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