

# (Z)-3-(o-Bromophenylamino)-2-nitrocrotonic acid methyl ester

**Inchi:** InChI=1S/C11H11BrN2O4/c1-7(10(14(16)17)11(15)18-2)13-9-6-4-3-5-8(9)12/h3-6,13H,1  
**InchiKey:** YTXNBNGNYHWPFP-YFHOEESVSA-N  
**Formula:** C11H11BrN2O4  
**SMILES:** COC(=O)C(=C(C)Nc1ccccc1Br)[N+](=O)[O-]  
**Mol. weight [g/mol]:** 315.12  
**CAS:** 127727-83-7

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 112.98  | kJ/mol  | Joback Method  |
| hf            | -123.43 | kJ/mol  | Joback Method  |
| hfus          | 40.01   | kJ/mol  | Joback Method  |
| hvap          | 81.75   | kJ/mol  | Joback Method  |
| log10ws       | -4.11   |         | Crippen Method |
| logp          | 2.542   |         | Crippen Method |
| mcvol         | 190.130 | ml/mol  | McGowan Method |
| pc            | 3243.03 | kPa     | Joback Method  |
| tb            | 831.12  | K       | Joback Method  |
| tc            | 1086.81 | K       | Joback Method  |
| tf            | 547.90  | K       | Joback Method  |
| vc            | 0.729   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 498.05 | J/mol×K | 831.12          | Joback Method |
| cpg           | 508.32 | J/mol×K | 873.74          | Joback Method |
| cpg           | 517.71 | J/mol×K | 916.35          | Joback Method |
| cpg           | 526.29 | J/mol×K | 958.97          | Joback Method |
| cpg           | 534.13 | J/mol×K | 1001.58         | Joback Method |
| cpg           | 541.32 | J/mol×K | 1044.20         | Joback Method |
| cpg           | 547.94 | J/mol×K | 1086.81         | Joback Method |

# Sources

|                        |                                                                                                                                                 |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                           |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C127727837&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C127727837&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                       |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                               |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                           |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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