

2,4-DIMETHYLDIPHENYLSULPHONE

Other names:	Sulfone, phenyl, 2,4-xylyl
Inchi:	InChI=1S/C14H14O2S/c1-11-8-9-14(12(2)10-11)17(15,16)13-6-4-3-5-7-13/h3-10H,1-2H3
InchiKey:	ZCVSDPUSEUOUHQ-UHFFFAOYSA-N
Formula:	C14H14O2S
SMILES:	Cc1ccc(S(=O)(=O)c2ccccc2)c(C)c1
Mol. weight [g/mol]:	246.33

Physical Properties

Property code	Value	Unit	Source
hf	-172.03	kJ/mol	Cheméo Relay (1.0)
hfus	30.70	kJ/mol	Joback Method
ie	8.55	eV	Cheméo Relay (1.0)
log10ws	-4.38		Cheméo Relay (1.0)
logp	3.136		Crippen Method
mcvol	188.690	ml/mol	McGowan Method
tb	641.75	K	Cheméo Relay (1.0)
tf	358.00 ± 2.00	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	456.66	J/mol×K	630.82	Joback Method
cpg	473.14	J/mol×K	669.29	Joback Method
cpg	488.39	J/mol×K	707.76	Joback Method
cpg	502.46	J/mol×K	746.22	Joback Method
cpg	515.36	J/mol×K	784.69	Joback Method
cpg	527.14	J/mol×K	823.16	Joback Method
cpg	537.83	J/mol×K	861.63	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C4212742&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point

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